

Atomic Structure Practice Worksheet

Part 1: The following table includes information about the name, symbol, atomic number, atomic mass, and the numbers of protons, electrons, and neutrons for several elements. Use the periodic table to fill in the blanks.

Element	Symbol	Atomic number	Atomic mass	Protons	Neutrons	Electrons
						3
Hydrogen						
		17				
	Au					
			30.974			
Tungsten						
				30		
				126.9		
		14				

Part 2: Use the periodic table to answer the following questions about atoms and ions.

① How many protons, neutrons, and electrons are present in $^{128}\text{Te}^{2-}$ ion?

Protons = _____ Neutrons = _____ Electrons = _____

② How many protons, neutrons, and electrons are present in $^{73}\text{Ge}^{4-}$ ion?

Protons = _____ Neutrons = _____ Electrons = _____

③ Write the complete chemical symbol for the ion with 82 protons and 80 electrons.

④ What is the atomic mass of an element that has 37 protons, 43 neutrons, and 36 electrons? _____

⑤ Write the complete chemical symbol for the ion with 73 protons, 108 neutrons, and 68 electrons? _____

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