

# Properties of Atoms and the Periodic Table

## Section 2 Masses of Atoms

**Preview** Section 2 of your book, using the checklist below.

Read all section titles.

- Read all boldfaced words.
- Read all charts and graphs.
- Look at all the pictures and read their captions.
- Think about what you already know about masses of atoms.

**Write three facts you learned.**

1. \_\_\_\_\_
2. \_\_\_\_\_
3. \_\_\_\_\_

### Review Vocabulary

**mass**

**Define** mass to show its scientific meaning.

A physical property of an object that measures its gravitational attraction to other bodies.

### New Vocabulary

**atomic number**

**Use your book or dictionary to define the following key terms.**

The number of protons in an atom's nucleus.

**mass number**

The sum of the number of protons and electrons in the nucleus of an atom.

**isotope**

Atoms of the same element that have different numbers of neutrons.

**average atomic mass**

The weighted average of mass of all naturally occurring isotopes of an element.

### Academic Vocabulary

**define**

**Use a dictionary to find the scientific meaning of define.**

To show or describe an element of science.

# Atoms And Their Properties Note Taking Sheets

**Annelies Wilder-Smith**

## Atoms And Their Properties Note Taking Sheets:

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