

	Isotope	Symbol	At. #	Mass #	# of p ⁺	# of n	# of e ⁻
a.	Aluminum-27	$^{27}_{13}\text{Al}$	13	27	13	14	13
b.	Calcium-39	$^{39}_{20}\text{Ca}$	20	39	20	19	20
c.	Sodium-23	$^{23}_{11}\text{Na}$	11	23	11	12	11
d.	Nickel-60	$^{60}_{28}\text{Ni}$	28	60	28	32	28
e.	Sulfur-34	$^{34}_{16}\text{S}$	16	34	16	18	16
f.	Argon-40	$^{40}_{18}\text{Ar}$	18	40	18	22	18
g.	Mercury-200	$^{200}_{80}\text{Hg}$	80	200	80	120	80
h.	Uranium-236	$^{236}_{92}\text{U}$	92	236	92	144	92

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