

	Isotope	Symbol	At. #	Mass #	# of p <sup>+</sup>	# of n	# of e <sup>-</sup>
a.	Aluminum-27	<sup>27</sup> <sub>13</sub> Al	13	27	13	14	13
b.	Calcium-39	<sup>39</sup> <sub>20</sub> Ca	20	39	20	19	20
c.	Sodium-23	<sup>23</sup> <sub>11</sub> Na	11	23	11	12	11
d.	Nickel-60	<sup>60</sup> <sub>28</sub> Ni	28	60	28	32	28
e.	Sulfur-34	<sup>34</sup> <sub>16</sub> S	16	34	16	18	16
f.	Argon-40	<sup>40</sup> <sub>18</sub> Ar	18	40	18	22	18
g.	Mercury-200	<sup>200</sup> <sub>80</sub> Hg	80	200	80	120	80
h.	Uranium-236	<sup>236</sup> <sub>92</sub> U	92	236	92	144	92

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**Kenneth S. Overway**



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