

2. An iron bar rusts. The iron reacts with oxygen in the air to make rust.



Which chemical formula(s) represents the *reactants*?



Which chemical formula(s) represents the *products*?



How many of each atom is present in the *reactants*?

Fe - 4 ✓
O - 6 ✓

How many of each atom is present in the *products*?

Fe - 4 ✓
O - 6 ✓

Is this a balanced equation? Explain.

Yes! The reactants = the products



Analyzing Chemical Equations Answer Key Only

Robert Mitchell



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General Chemistry with Qualitative Analysis Ralph H. Petrucci, Robert K. Wismer, 1987 **Journal of Research of the National Bureau of Standards** United States. National Bureau of Standards, 1976 Paint Analysis Roger Dietrich, 2021-04-12 The market demands modern high performance flawless paints that possess specified properties Where deviations from set points occur the cause must be investigated and the error must be remedied What standard methods don't disclose is why a particular coating either meets or fails to meet a requirement Thus the author presents modern analytical techniques and their applications in the coatings industry that answer further complex questions The information in this book can be used for performing failure analysis production control and quality control and also meet the requirements of modern high level quality management An excellent combination of theory and practice for formulators paint engineers and applied technologists seeking a sound basic introduction to instrumental paint analysis and concrete answers to everyday problems

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Analysis of Essential Nuclear Reactor Materials U.S. Atomic Energy Commission. Laboratory, New Brunswick, N.J., 1964

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