

Name \_\_\_\_\_

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### Percent Yield Practice Problems

**Directions:** Determine the percent yield for each scenario in the problems below. You may need to determine the theoretical yield using stoichiometry before determining the percent yield.

$$\text{Percent Yield} = \frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100$$

1. Determine the percent yield if a reaction yielded 23.5 grams of water vapor. The theoretical yield of the reaction was 32.0 grams of water vapor.

the percent yield is 73.44%

2. A student performs an experiment and generates 5.0 grams of hydrogen gas. They calculate the theoretical yield of their reaction and determine it should have generated 7.5 grams of hydrogen gas. What was the percent yield for the student's experiment?

the percent yield is 66.67%

3. An industrial-grade reaction generates 4,500 kg of product. The company anticipated a yield of 5,500 kg. What is the percent yield for this reaction?

the percent yield is 81.82%

4. A pharmaceutical drug company has a prescription drug that has a percent yield of 76%. If the company uses enough materials to theoretically generate 8,000 kg of product, how much will they actually produce?

6,080 kg

## 12 3 Practice Problems Percent Yield

**Larry Joel Goldstein,David C.  
Lay,David I. Schneider**

## 12 3 Practice Problems Percent Yield:

Calculating Theoretical Yield Is Not Impossible! Michael S. Leonard,2025-02-25 Calculating theoretical and percent yield is a fundamental skill for the laboratory This book primarily targets Organic Chemistry Laboratory courses at the high school or college and university level as a supplemental resource to help students master this skill It begins with simple examples from everyday life demonstrates the importance of balancing the equation addresses the role of the mole in these computations discusses different types of liquids considers the role of significant figures and culminates with the planning of syntheses There are suggestions for further reading as well as practice problems and questions to ensure mastery Begins with examples from everyday life that enable students to understand the concepts of theoretical and percent yield before applying those concepts to the laboratory Addresses the necessity of balancing the reaction equation the centrality of the mole in these calculations and the role of significant figures in reporting the answer Explains how to approach the calculations when using neat liquids or solutions The culmination of this text is the use of the same thought processes to plan the amounts of reactants needed for syntheses of desired quantities of product All of the problems in the book include detailed solutions with accompanying text to explain the answers and ancillaries also include suggestions for further reading

**Materials Performance and the Deep Sea** ,1969      **Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook** McGraw Hill,2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change      **General Organic and Biological Chemistry** Kenneth W. Raymond,2009-12-14 This general organic and biochemistry text has been written for students preparing for careers in health related fields such as nursing dental hygiene nutrition medical technology and occupational therapy It is also suited for students majoring in other fields where it is important to have an understanding of the basics of chemistry Students need have no previous background in chemistry but should possess basic math skills The text features numerous helpful problems and learning features      Holt Chemistry Ralph Thomas Myers,2004      *Materials Performance and the Deep Sea* ,1969      **Partial Solutions Manual** Darrell D. Ebbing,David Bookin,1998      **Chemical Engineering Reference Manual** Randall N. Robinson,1996 The Chemical Engineering Reference Manual provides a detailed review for engineers studying for the chemical PE exam preparing them for what they will find on test day It includes more than 160 solved example problems 164 practice problems and test taking strategy The chemical PE exam is an eight hour open book test consisting of 80 multiple choice problems It is administered every April and October The Chemical Engineering Reference Manual is the primary text examinees need both to prepare for and to use during the exam It reviews current exam topics and uses practice problems to emphasize key concepts Supplementary products include the Solutions Manual for the practice problems and the Practice PE Exams      *Proceedings of the Sixteenth Annual Conference of the Cognitive Science Society* Ashwin Ram,Kurt Eiselt,2019-05-23 This volume features the complete text of all regular papers posters and summaries of symposia presented at the 16th annual meeting of

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