

PERCENT COMPOSITION

Name _____

Class _____

Date _____

For each chemical formula below, calculate the percent (by mass) composition of each element. Start by calculating the molar mass and then calculate the percent by mass. Round to the nearest hundredth.

Percent Composition from Chemical Formula

1. H_2S

Molar Mass Calculation:

$$H: (1.008) \times 2 = 2.016$$

$$S: (32.06) \times 1 = 32.06$$

$$H_2S \quad 34.076 \text{ g/mole}$$

Percent Composition:

$$H: \frac{2.016}{34.076} \times 100 = \frac{2.016}{34.076} = 5.92\%$$

$$S: \frac{32.06}{34.076} \times 100 = \frac{32.06}{34.076} = 94.08\%$$

2. SO_2

Molar Mass Calculation:

Percent Composition:

3. H_2SO_4

Molar Mass Calculation:

Percent Composition:

4. HCl

Molar Mass Calculation:

CHEMISTRY HOMEWORK PAGE

103 Percent Composition And Chemical Formulas Answer Key

Elizabeth P. Rogers



103 Percent Composition And Chemical Formulas Answer Key:

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