

THE ANSWER

KEY

Volume Two

$$\frac{2.88 \text{ grams H}_2\text{O}}{1} \times \frac{1 \text{ moles H}_2\text{O}}{18.00 \text{ grams H}_2\text{O}} \times \frac{1 \text{ moles O}}{1 \text{ moles H}_2\text{O}} = 2.56 \text{ grams O}$$

A Comprehensive Explanation of Problem Solving Methods
for General Chemistry Success

RACHEL TUROSCY

Answer Key To Problem Solving In Chemistry

Dorothy Gabel

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