

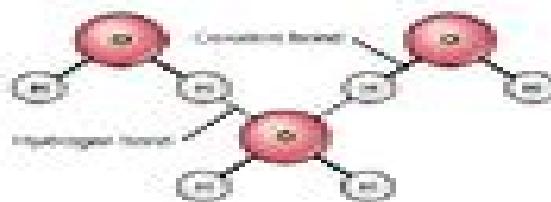
Unit 1 study guide

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| Created | September 10, 2007 2:25 PM |
| Class | Unit 1: Chemistry of Life |
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Topic 3.1 : Structure of Water and Hydrogen Bonding

1. Structure & Polarity

- 2 elements : Oxygen and Hydrogen (1:2 ratio)
- Oxygen is more electronegative than hydrogen, resulting in an unequal sharing of electrons between oxygen and hydrogen. This is called Polarity, where there are differences in atoms electronegativity, a water molecule has polarity.
- The bond is a covalent bond between the two elements (they share electrons)
- The bond between other water molecules are hydrogen bonds (a weak bond between the negative and positive regions of two separate molecules).



2. Properties of water

LIVING THINGS DEPEND ON WATER'S PROPERTIES

- When two of the same molecules form hydrogen bonds with each other this is called cohesion (water and water).
- When two different molecules form hydrogen bonds with each other this is called adhesion (amino acid and water).

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**Paul Anthony Samuelson,George B.
Baldwin**

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CISSP Certification Exam Study Guide Kumud Kumar,2023-07-17 This book has been carefully crafted to delve into each of the 8 CISSP Common Body of Knowledge CBK domains with comprehensive detail ensuring that you gain a solid grasp of the content The book consists of 8 chapters that form its core Here s a breakdown of the domains and the chapters they are covered in Chapter 1 Security and Risk Management Chapter 2 Asset Security Chapter 3 Security Architecture and Engineering Chapter 4 Communication and Network Security Chapter 5 Identity and Access Management IAM Chapter 6 Security Assessment and Testing Chapter 7 Security Operations Chapter 8 Software Development Security This book includes important resources to aid your exam preparation such as exam essentials key terms and review questions The exam essentials highlight crucial topics that you should focus on for the exam Throughout the chapters you will come across specialized terminology which is also conveniently defined in the glossary at the end of the book Additionally review questions are provided to assess your understanding and retention of the chapter s content

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