

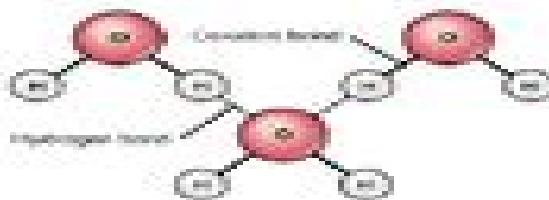
Unit 1 study guide

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Class	Unit 1: Chemistry of Life
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Topic 3.1 : Structure of Water and Hydrogen Bonding

1. Structure & Polarity

- 2 elements : Oxygen and Hydrogen (1:2 ratio)
- Oxygen is more electronegative than hydrogen, resulting in an unequal sharing of electrons between oxygen and hydrogen. This is called Polarity, where there are differences in atoms electronegativity, a water molecule has polarity.
- The bond is a covalent bond between the two elements (they share electrons)
- The bond between other water molecules are hydrogen bonds (a weak bond between the negative and positive regions of two separate molecules).



2. Properties of water

LIVING THINGS DEPEND ON WATER'S PROPERTIES

- When two of the same molecules form hydrogen bonds with each other this is called cohesion (water and water).
- When two different molecules form hydrogen bonds with each other this is called adhesion (amino acid and water).

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**Paul Anthony Samuelson,George B.
Baldwin**

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tout savoir sur le jour de la terre en 4 questions - Feb 17 2023

apr 22 2022 le jour de la terre c est surtout un réseau international d initiatives pour dire haut et fort que l on tient à notre planète tout le monde peut participer et célébrer ce jour à sa façon avec sa famille ses amis ses collègues ses voisins ou seul