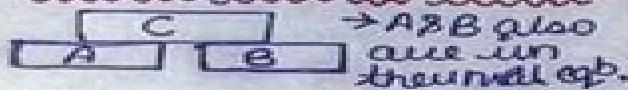


branch of science with conversion of heat and work.

\* Zeroth Law of Thermodynamics



\* First Law of Thermodynamics  
 $\rightarrow dQ = dw + \Delta U$

• work done  $\rightarrow dw = \int P dV$   
 • if  $P$  const;  $w = P(V_2 - V_1)$



$$\Delta U = \frac{f}{2} nR\Delta T$$

$$\{ \Delta U = nC_V\Delta T \}$$

$$\{ \Delta U = \frac{nR\Delta T}{Y-1} \} \frac{f}{2} = \frac{1}{Y-1}$$

$$\{ \Delta U = \frac{P_2V_2 - P_1V_1}{Y-1} \}$$

\* Sign Conventions

→ Heat in  $\rightarrow$  +ve out  $\rightarrow$  -ve

$$\Delta U = \frac{f}{2} nR\Delta T$$

$$T \uparrow \rightarrow \Delta T \uparrow \therefore \Delta U \rightarrow +ve$$

$$T \downarrow \rightarrow \Delta T \downarrow \therefore \Delta U = -ve$$

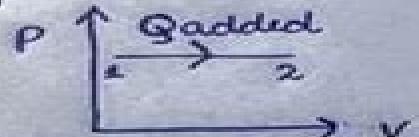
Q:  $\int_{100J}^{100J} = 0$  Find  $\Delta U = ?$   
 $\rightarrow w = 60J$   $dQ = dw + \Delta U$   
 $\Delta U = 40J$

$$Q_{out} = -ve \quad Q_{in} = +ve$$

(i) Isobaric  $\therefore (P=C)$

$$PV = nRT \rightarrow V \propto T$$

$$\sqrt{T} = \text{constant}$$



## Thermodynamics

$$w = \int P \cdot dV \rightarrow w = P(V_2 - V_1)$$

$$\Delta U = nC_V\Delta T \text{ you can prove}$$

$$dw = nC_P\Delta T$$

(ii) Isochoric  $\rightarrow$   $V = \text{const}$

$$P \propto T \rightarrow P/T = \text{constant}$$

$\rightarrow$  Gay-Lussac law.

$$\{ \Delta U = 0 \}$$

$$\Delta U = nC_V\Delta T$$

$$dQ = nC_V\Delta T$$

(iii) Isothermal (Boyle's law)

$$PV = \text{constant}, \quad XY = \text{constant}$$

$$w = P_1V_1 \ln \frac{V_2}{V_1}$$

$$\Delta U = 0 \quad w = nRT \ln \frac{V_2}{V_1}$$

$$dQ = dw$$

(iv) Adiabatic  $Q=0$

$\rightarrow$  Bursting of tyres

$$PV^Y = \text{constant}$$

$$\{ Q=0 \}$$

$$\text{mono} = 1.66$$

$$\text{dia} = 1.4$$

$$\text{Poly} = 1.33$$

$$dw = -\Delta U$$

$$dQ = nR(T_2 - T_1)$$

$$\Delta U = nC_V\Delta T \quad Y-1$$

$$\frac{P_2}{P_1} = \left( \frac{V_1}{V_2} \right)^Y, \quad \frac{T_2}{T_1} = \left( \frac{P_2}{P_1} \right)^{\frac{1}{Y-1}}$$

$$\frac{T_2}{T_1} = \left( \frac{V_1}{V_2} \right)^{Y-1}$$

$$(\text{slope})_{\text{baro}} = Y (\text{slope})_{\text{isotherm}}$$

$$\text{isotherm}$$

$$\text{adiabatic}$$

(v) Polytropic  $\therefore$

$$PV^n = \text{constant}$$

$$Q \neq 0$$

$$dQ = dw + \Delta U$$

$$dQ = d\omega + \Delta U$$

$$d\omega = nR(T_2 - T_1)$$

$$n-1$$

$$\Delta U = nRAT$$

$$Y-1$$

$$dQ = nC_V\Delta T$$

$$P = \text{constant}$$

$$PV = \text{constant}$$

$$PV^Y = \text{constant}$$

$$V = \text{constant}$$

$$P = C$$

$$n=0$$

$$iso \rightarrow n=1$$

$$adi \rightarrow n=\infty$$

$$vol \rightarrow \infty$$

$$P \rightarrow 0$$

$$n \rightarrow \infty$$

$$cyclic$$

$$process$$

$$exa$$

$$area enclosed$$

$$\Delta U = \text{Point}$$

$$\omega = \text{Path}$$

$$V \quad Q = \text{Path}$$

$$Point \text{ and Path}$$

$$T_1$$

$$T_2$$

$$T_3$$

$$T_4$$

$$T_5$$

$$T_6$$

$$T_7$$

$$T_8$$

$$T_9$$

$$T_{10}$$

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