

Guided Reading

Periodic Table

What is the Periodic Table?

Periodic Table: Organization

Periodic table organization emphasizes
atomic theory principles.

Three Major Trends in Periodic Table

Periodic Table organization: Three major trends: 1) increasing atomic number from left to right across a period; 2) increasing atomic number from bottom to top in a group; 3) increasing atomic number from left to right in a block.

Three major trends in periodic table organization are: increasing atomic number from left to right across a period; increasing atomic number from bottom to top in a group; increasing atomic number from left to right in a block.

1. As we move from left to right, atomic properties are like **Progressive Changes**: these transitions involving increasing atomic number are called **Periodic Transitions**.

2. The vertical columns are called the **Groups** (or families) of elements. A group contains four or six elements.

3. Horizontal rows are called the **Periods** (or main groups).

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