

Common Polyatomic Ions

Formula	Name	Formula	Name
NH_4^+	Ammonium	CrO_4^{2-}	Chromate
NH_3	Ammonia	$\text{Cr}_2\text{O}_7^{2-}$	Dichromate
$\text{C}_2\text{H}_3\text{O}_2^-$	Acetate	MnO_4^-	Permanganate
CH_3COO^-	Acetate	MnO_4^{2-}	Manganate
CN^-	Cyanide	NO_2^-	Nitrite
CO_3^{2-}	Carbonate	NO_3^-	Nitrate
HCO_3^-	Bicarbonate	OH^-	Hydroxide
$\text{C}_2\text{O}_4^{2-}$	Oxalate	PO_4^{3-}	Phosphate
ClO^-	Hypochlorite	SCN^-	Thiocyanate
ClO_2^-	Chlorite	$\text{Fe}(\text{CN})_6^{3-}$	Ferricyanide

ClO_3^-	Chlorate	SO_3^{2-}	Sulfite
ClO_4^-	Perchlorate	SO_4^{2-}	Sulfate
$\text{S}_2\text{O}_3^{2-}$	Thiosulfate	HSO_4^-	Hydrogen sulfate
BrO^-	Hypobromite	IO_3^-	Iodate
AsO_2^{3-}	Arsenite	SeO_4^{2-}	Selenate
BrO_3^-	Bromate	HSO_3^-	Hydrogen

17 Polyatomic Ions S Answer Key

Joan C. Sauerwein



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Captures the current fascination with forensics through Chemistry at the Crime Scene boxed sections For those interested in basic chemistry **Study Guide, Chemical Principles, Fifth Edition, Zumdahl** Paul B. Kelter, Steven S. Zumdahl, 2004-04 **INIS Atomindeks** ,1986 **Chemical Principles, Properties, and Reactions** Kenneth L. Henold, Frank Walmsley, 1984 Chemistry Charles H. Corwin, 1994 The book focuses on the concepts of chemistry and the applications that maintain and generate motivation for the subject of chemistry

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