

CHEM 1406 Practice Final Exam

1. How many significant figures should there be in the answer to the following problem?

$$\frac{(3.3 - 20.0) \times (1.79 \times 10^3)}{139}$$

- A) 1
- B) 2
- ☒ C) 3
- D) 4
- E) none of the above

2. A nurse must administer the appropriate dosage of atropine sulfate to a 165-lb male patient. The dosage on the label of a bottle of atropine sulfate tablets reads 20 mg for every kilogram of body weight. If each tablet contains 0.5 grams of atropine sulfate, how many tablets are needed to give the correct dose to the patient?

- A) half tablet
- B) 1 tablet
- C) two tablets
- ☒ D) three tablets
- E) none of the above

3. Which of the following is the largest mass?

- A) $4.1 \cdot 10^{-6}$ kg
- ☒ B) 22 g
- C) $8.2 \cdot 10^9$ ng
- D) $3 \cdot 10^3$ mg
- E) 352 cg

4. Tequila has a density of 0.78 g/mL. What mass of Tequila will fill a 2.0 fluid ounce shot glass? (29.6 mL = 1 liquid ounce)

- A) $4.6 \cdot 10^{-6}$ kg
- ☒ B) 46 g
- C) $8.2 \cdot 10^6$ μ g
- D) $3 \cdot 10^3$ mg
- E) none of the above

5. The melting point of a substance is an example of a(n)...

- ☒ A) Physical property
- B) Chemical property
- C) Specific heat
- D) Extensive property
- E) none of the above

1406 Final Test Answer

John Kerr, Holly Robertson, Leckie

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