

## Types of Chemical Reactions Lab Answer Key

### Chemical Reaction #1

<b>Reactants</b> (write in words)	hydrochloric acid and magnesium
<b>Observations</b>	accept most responses
<b>Type of Reaction</b>	single replacement
<b>Predict Products</b> (write in words)	magnesium chloride and hydrogen gas
<b>Equation</b> (write in symbols)	$2\text{HCl}_{\text{aq}} + \text{Mg}_{\text{ss}} \rightarrow \text{MgCl}_{\text{aq}} + \text{H}_{\text{g}}$

### Chemical Reaction #2

<b>Reactants</b> (write in words)	copper metal and oxygen gas
<b>Observations</b>	accept most responses
<b>Type of Reaction</b>	synthesis
<b>Predict Products</b> (write in words)	copper oxide
<b>Equation</b> (write in symbols)	$2\text{Cu}_{\text{ss}} + \text{O}_{\text{g}} \rightarrow 2\text{CuO}_{\text{ss}}$

### Chemical Reaction #3

<b>Reactants</b> (write in words)	methane gas and oxygen gas
<b>Observations</b>	accept most responses
<b>Type of Reaction</b>	combustion
<b>Predict Products</b> (write in words)	carbon dioxide and water
<b>Equation</b> (write in symbols)	$\text{CH}_{\text{4g}} + 2\text{O}_{\text{2g}} \rightarrow \text{CO}_{\text{2g}} + 2\text{H}_2\text{O}_{\text{2g}}$

### Chemical Reaction #4

<b>Reactants</b> (write in words)	hydrogen peroxide
<b>Observations</b>	accept most responses
<b>Type of Reaction</b>	decomposition
<b>Predict Products</b> (write in words)	water and oxygen gas
<b>Equation</b> (write in symbols)	$2\text{H}_2\text{O}_{\text{2l}} \rightarrow 2\text{H}_2\text{O}_{\text{2g}} + \text{O}_{\text{2g}}$

# Answer Key To Lab On Chemical Reactions

**Brian Pressley**

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