

NAME: \_\_\_\_\_

## Getting to Know the Elements!

Use your Periodic Table and note pages to help you get to know these elements a little better! Remember...APE MAN!

1. How many protons does Calcium have? \_\_\_\_\_
2. How many electrons does Yttrium have? \_\_\_\_\_
3. How many neutrons does Titanium have? \_\_\_\_\_
4. The sub atomic particle that has a neutral charge is the: \_\_\_\_\_
5. The sub atomic particle that has a negative charge is the: \_\_\_\_\_
6. The sub atomic particle that has a positive charge is the: \_\_\_\_\_
7. What element has 13 protons? \_\_\_\_\_
8. What element has 2 neutrons? \_\_\_\_\_
9. What element has 0 neutrons? \_\_\_\_\_
10. How many electrons does Cm have? \_\_\_\_\_
11. What is the sum of the atomic masses of Bi and Br? \_\_\_\_\_
12. How many neutrons does Gold have? \_\_\_\_\_
13. What is the sum of the number of protons of Iridium and the number of protons in Strontium? \_\_\_\_\_
14. What is the element symbol for Platinum? \_\_\_\_\_
15. What's the difference between the number of electrons in an atom of Tungsten and the electrons in an atom of Tin? \_\_\_\_\_
16. What particles are found in the nucleus of an atom? \_\_\_\_\_ and \_\_\_\_\_
17. Where do you find the electrons in an atom? \_\_\_\_\_
18. How many electrons can the 2<sup>nd</sup> electron cloud hold? \_\_\_\_\_

# Answer Key I Getting To Know

**George Ehrenhaft**



## **Answer Key I Getting To Know:**

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