

CHEMISTRY: COUNTING ATOMS IN COMPOUNDS WORKSHEET #7.0.1

INSTRUCTIONS: Write the quantity of atoms of each element opposite the formula of the compound for the quantity of formula units and molecules shown:

For example: $5\text{P}_2\text{O}_3$ $\text{P} = \underline{(5 \times 2 =) 10}$ $\text{O} = \underline{(5 \times 3 =) 15}$

For example: $4\text{Zn}(\text{NO}_3)_2$ $\text{Zn} = \underline{(4 \times 1 =) 4}$ $\text{N} = \underline{(4 \times 1 \times 2 =) 8}$ $\text{O} = \underline{(4 \times 3 \times 2 =) 24}$

1. $4\text{K}_2\text{CO}_3$ $\text{K} = \underline{8}$ $\text{C} = \underline{4}$ $\text{O} = \underline{12}$
2. $2\text{Sr}_3(\text{PO}_4)_2$ $\text{Sr} = \underline{6}$ $\text{P} = \underline{4}$ $\text{O} = \underline{16}$
3. $3\text{N}_4\text{O}_{10}$ $\text{N} = \underline{12}$ $\text{O} = \underline{30}$
4. $2(\text{NH}_4)_3\text{N}$ $\text{N} = \underline{8}$ $\text{H} = \underline{24}$
5. $8\text{Cl}_2\text{O}$ $\text{Cl} = \underline{16}$ $\text{O} = \underline{8}$
6. $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ $\text{Ca} = \underline{1}$ $\text{C} = \underline{4}$ $\text{H} = \underline{6}$ $\text{O} = \underline{4}$
7. 12NaBr $\text{Na} = \underline{12}$ $\text{Br} = \underline{12}$

Answer Key For Counting Atoms In Compounds

Kenneth W. Raymond



Answer Key For Counting Atoms In Compounds:

Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant Revise As and A2 - Chemistry Rob Ritchie, 2008-10 Revise AS A2 Chemistry gives complete study support throughout the two A Level years This Study Guide matches the curriculum content and provides in depth course coverage plus invaluable advice on how to get the best results in the exams **E3 Chemistry Guided Study Book - 2018 Home Edition (Answer Key Included)** Effiong Eyo, 2017-12-08 Chemistry students and Homeschoolers Go beyond just passing Enhance your understanding of chemistry and get higher marks on homework quizzes tests and the regents exam with E3 Chemistry Guided Study Book 2018 With E3 Chemistry Guided Study Book students will get clean clear engaging exciting and easy to understand high school chemistry concepts with emphasis on New York State Regents Chemistry the Physical Setting Easy to read format to help students easily remember key and must know chemistry materials Several example problems with guided step by step solutions to study and follow Practice multiple choice and short answer questions along side each concept to immediately test student understanding of the concept 12 topics of Regents question sets and 2 most recent Regents exams to practice and prep for any Regents Exam This is the Home Edition of the book Also available in School Edition ISBN 978 1979088374 The Home Edition contains answer key to all questions in the book Teachers who want to recommend our Guided Study Book to their students should recommend the Home Edition Students and and parents whose school is not using the Guided Study Book as instructional material as well as homeschoolers should also buy the Home edition The School Edition does not have the answer key in the book A separate answer key booklet is provided to teachers with a class order of the book Whether you are using the school or Home Edition our E3 Chemistry Guided Study Book makes a great supplemental instructional and test prep resource that can be used from the beginning to the end of the school year PLEASE NOTE Although reading contents in both the school and home editions are identical there are slight differences in question numbers choices and pages between the two editions Students whose school is using the Guided Study Book as instructional material SHOULD NOT buy the Home Edition Also available in paperback print *Chemistry insights 'O' level* Rex M. Heyworth, 2007 **Introductory Chemistry** Nivaldo J. Tro, 2006 **Sif: Chemistry 5na Tb** J. G. R. Briggs, 2009 *The Chemical Element* Javier García-Martínez, Elena Serrano-Torregrosa, 2011-09-19 In the International Year of Chemistry prominent scientists highlight the major advances in the fight against the largest problems faced by

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