

CHEMISTRY: COUNTING ATOMS IN COMPOUNDS WORKSHEET #7.0.1

INSTRUCTIONS: Write the quantity of atoms of each element opposite the formula of the compound for the quantity of formula units and molecules shown:

For example: $5\text{P}_2\text{O}_5$ $\text{P} = \underline{(5 \times 2 =) 10}$ $\text{O} = \underline{(5 \times 5 =) 25}$

For example: $4\text{Zn}(\text{NO}_3)_2$ $\text{Zn} = \underline{(4 \times 1 =) 4}$ $\text{N} = \underline{(4 \times 1 \times 2 =) 8}$ $\text{O} = \underline{(4 \times 3 \times 2 =) 24}$



Answer Key For Counting Atoms In Compounds

Kenneth W. Raymond

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Chemistry John A. Olmsted, Gregory M. Williams, Robert Charles Burk, 2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind. A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings. It more accurately reflects the curriculum of most Canadian institutions. Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant. [Revise AS and A2 - Chemistry](#) Rob Ritchie, 2008-10 Revise AS A2 Chemistry gives complete study support throughout the two A Level years. This Study Guide matches the curriculum content and provides in depth course coverage plus invaluable advice on how to get the best results in the exams.

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1e Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry **Matter**,

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