

**General Instructions :**

Read the following instructions very carefully and strictly follow them :

- (i) This question paper comprises **15** questions. **All** questions are compulsory.
- (ii) This question paper is divided into **three** sections – **A, B** and **C**.
- (iii) **Section A** – Questions No. **1** to **7** are short answer type questions. Each question carries **2** marks.
- (iv) **Section B** – Questions No. **8** to **13** are also short answer type questions. Each question carries **3** marks.
- (v) **Section C** – Questions No. **14** and **15** are case-based questions. Each question carries **4** marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

**SECTION A**

- 1. (a) Write the molecular formula of the following carbon compounds :
  - (i) Methane
  - (ii) Propane(b) Carbon compounds have low melting and boiling points. Why ? 2
- 2. The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.
  - (a) Write the group numbers of these elements in the Modern Periodic Table.
  - (b) Write the molecular formula of the compound formed when X and Y combine with each other.
  - (c) Which of the two is electropositive ? 2
- 3. (a) Which of the following flowers will have higher possibility of self-pollination ?  
Mustard, Papaya, Watermelon, Hibiscus  
(b) List the two reproductive parts of a bisexual flower. 2
- 4. Which one of the two multicellular organisms — Spirogyra and Planaria reproduces by regeneration and why ? Give an example of any other organism which can also reproduce by the same process. 2



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**Rajah Bhushanam Manikam**



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## **Table of Contents 2physical Science Grade Question Paper**

1. Understanding the eBook 2physical Science Grade Question Paper
  - The Rise of Digital Reading 2physical Science Grade Question Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying 2physical Science Grade Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 2physical Science Grade Question Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from 2physical Science Grade Question Paper
  - Personalized Recommendations
  - 2physical Science Grade Question Paper User Reviews and Ratings
  - 2physical Science Grade Question Paper and Bestseller Lists
5. Accessing 2physical Science Grade Question Paper Free and Paid eBooks
  - 2physical Science Grade Question Paper Public Domain eBooks
  - 2physical Science Grade Question Paper eBook Subscription Services
  - 2physical Science Grade Question Paper Budget-Friendly Options

6. Navigating 2physical Science Grade Question Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - 2physical Science Grade Question Paper Compatibility with Devices
  - 2physical Science Grade Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of 2physical Science Grade Question Paper
  - Highlighting and Note-Taking 2physical Science Grade Question Paper
  - Interactive Elements 2physical Science Grade Question Paper
8. Staying Engaged with 2physical Science Grade Question Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers 2physical Science Grade Question Paper
9. Balancing eBooks and Physical Books 2physical Science Grade Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection 2physical Science Grade Question Paper
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine 2physical Science Grade Question Paper
  - Setting Reading Goals 2physical Science Grade Question Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 2physical Science Grade Question Paper
  - Fact-Checking eBook Content of 2physical Science Grade Question Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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