

30. A scientific law is established only after

- discoveries have been made.
- experiments have been carried out and proven true.
- hypothesis has been tested.
- reasonable explanations have been put forward.
- theory has been extensively tested and proven true.

31. The following are properties of group VI elements **except** that their

- atomic radius increases down the group.
- electronegativity decreases down the group.
- melting and boiling points increase down the group.
- metallic properties decrease down the group.
- outer most shell has six electrons

32. What volume in dm^3 of water will be added to 1dm^3 of 6 moldm^{-3} solution to make it 2 moldm^{-3} ?

- 0.2
- 0.3
- 2.0
- 3.0
- 12.0

33. In the periodic table, elements in the same period show

- progressive decrease in electronegativity.
- progressive increase in atomic number.
- similarities in chemical properties.
- similarities in electronic structures.
- similarities in physical properties.

34. If the rate of diffusion of oxygen is $7.5\text{ cm}^3/\text{s}$ and the time taken to diffuse is 12 seconds, calculate the volume in (cm^3).

- 80
- 89
- 90
- 91
- 99

35. What similarity exists between a chlorine atom and chlorine ion?

- Chemical property
- Electron charge
- Electronic configuration
- Number of electrons
- Number of protons

36. 28.4 g of a salt dissolved in 250 cm^3 of distilled water to give a concentration of 0.8 moldm^{-3} . Calculate the molar mass of the salt.

- 35.5
- 71.0
- 142.0
- 250.0
- 278.4

2014 Chemistry Objective Answers

Deepak Mudgil, Sheweta Barak Mudgil

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