

30. A scientific law is established only after
- discoveries have been made.
  - experiments have been carried out and proven true.
  - hypothesis has been tested.
  - reasonable explanations have been put forward.
  - theory has been extensively tested and proven true.
31. The following are properties of group VI elements **except** that their
- atomic radius increases down the group.
  - electronegativity decreases down the group.
  - melting and boiling points increase down the group.
  - metallic properties decrease down the group.
  - outer most shell has six electrons
32. What volume in  $\text{dm}^3$  of water will be added to  $1\text{dm}^3$  of  $6\text{mol dm}^{-3}$  solution to make it  $2\text{mol dm}^{-3}$ ?
- 0.2
  - 0.3
  - 2.0
  - 3.0
  - 12.0
33. In the periodic table, elements in the same period show
- progressive decrease in electronegativity.
  - progressive increase in atomic number.
  - similarities in chemical properties.
  - similarities in electronic structures.
  - similarities in physical properties.
34. If the rate of diffusion of oxygen is  $7.5\text{ cm}^3/\text{s}$  and the time taken to diffuse is 12 seconds, calculate the volume in ( $\text{cm}^3$ ).
- 80
  - 89
  - 90
  - 91
  - 99
35. What similarity exists between a chlorine atom and chlorine ion?
- Chemical property
  - Electron charge
  - Electronic configuration
  - Number of electrons
  - Number of protons
36. 28.4 g of a salt dissolved in  $250\text{ cm}^3$  of distilled water to give a concentration of  $0.8\text{mol dm}^{-3}$ . Calculate the molar mass of the salt.
- 35.5
  - 71.0
  - 142.0
  - 250.0
  - 278.4

# 2014 Chemistry Objective Answers

**Deepak Mudgil, Sheweta Barak Mudgil**

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