

Advanced Placement Chemistry: 1992 Free Response Answers

- [delta] and [sigma] are used to indicate the capital Greek letters.
- [square root] applies to the numbers enclosed in parentheses immediately following.
- All simplifying assumptions are justified within 5%.
- One point deduction for a significant figure or math error, applied only once per problem.
- No credit earned for numerical answer without justification.

1) average points = 3.0

a) three points

mole fraction is 50:50 from equation

therefore $7.76 \text{ atm} = 2 = 3.88 \text{ atm}$ (contribution by H_2O)

$(P^\circ) = \text{atm}$; rearrange to get $n = (P^\circ) \cdot (RT)$

$n = [(3.88 \text{ atm}) (0.0821 \text{ L atm mol}^{-1} \text{ K}^{-1}) (403 \text{ K})]$
 $= 0.547 \text{ mol}$

Personal note by John Park: the actual AP standard uses 7.76 atm in its $P^\circ = \text{atm}$ calculation and then applies the 50:50 criterion to the value for n (0.59-mol) calculated. I prefer the above explanation.

b) two points

from equation, the ratio of NaHCO_3 to H_2O is 2:1

$n = 0.547 = 2 \cdot x$

$x = 0.274 \text{ mol of NaHCO}_3 \text{ decomposed}$

$0.274 \text{ mol} \times 84.01 \text{ g/mol} = 23.0 \text{ g of NaHCO}_3 \text{ decomposed}$

$100. \text{ g} - 23.0 \text{ g} = 77.0 \text{ g of NaHCO}_3 \text{ remaining}$

c) two points

$K_p = (P_{\text{H}_2\text{O}}) (P_{\text{CO}_2})$

$= (3.88 \text{ atm}) (3.88 \text{ atm}) = 15.0 \text{ atm}^2$

d) two points

The total pressure would remain at 7.76 atm . Since some solid remained when $100. \text{ g}$ was used (and there had been no temperature change), then using $100. \text{ g}$ would not affect the equilibrium.

1) average points = 4.0

a) three points

$2.50 \text{ amps} = 2.50 \text{ C/sec}$; $35.0 \text{ min} = 2.10 \times 10^3 \text{ sec}$

$(2.50 \text{ C/sec}) (2.10 \times 10^3 \text{ sec}) = 5.25 \times 10^3 \text{ Coulombs delivered}$

$(5.25 \times 10^3 \text{ C}) = (1 \text{ Faraday} / 96,500 \text{ C}) = 0.0544 \text{ mol } e^- \text{ delivered}$

since $\text{M}^{2+} + 2e^- \rightarrow \text{M}$, therefore $0.0544 \div 2 = 0.0272 \text{ mol M}^{2+} \text{ deposited}$

$3.00 \text{ g} = 0.0272 \text{ mol} = 111.5 \text{ g/mol}$

From an examination of the periodic table, the metal is seen to be cadmium.

b) two points

$\text{Cd}^{2+} + 2e^- \rightarrow \text{Cd}$

$E^\circ = -0.40 \text{ V}$

$\text{Zn} \rightarrow \text{Zn}^{2+} + 2e^-$

$E^\circ = +0.76 \text{ V}$

$\text{Cd}^{2+} + \text{Zn} \rightarrow \text{Cd} + \text{Zn}^{2+}$

$E^\circ = 0.36 \text{ V}$

c) two points

$[\text{delta}]G^\circ = -FE^\circ$

$= -(2 \text{ mol}) (96,500 \text{ C/mol}) (0.36 \text{ V})$ [see also the note in (1) (b) (c) (c)]

$= -69,800 \text{ J} = -69.8 \text{ kJ}$ (after appropriate rounding off)

d) two points

$E = E^\circ - (0.0591 / n) \log ([\text{Zn}^{2+}] / [\text{Cd}^{2+}])$

$E = 0.36 - (0.0591 / 2) \log (0.01 / 1)$

$E = 0.36 - (-0.0591) = 0.39 \text{ V}$

14 Chemistry Response Answers

**Hyoung Joon Choi, Takhee Lee, Woo-
Sung Jung**



14 Chemistry Response Answers:

Journal of Chemical Education, 1928 Career Point Kota JEE Main 2021 Chapterwise Solved Papers Physics, Chemistry, and Mathematics Career Point Kota, 2022-03-27 Here s introducing the all new edition of 2021 JEE Main Chapterwise Solved Papers this book has been comprehensively comprised of all 26 Sets of online papers that were conducted in February March July and August Giving complete detailed and authentic solutions to all the questions this book serves as a must have practice manual before the final call in the examination hall Whenever a student decides to prepare for any examination her his first and foremost curiosity about the type of questions that he she has to face This becomes more important in the context of competitive examinations where there is neck to neck race We feel great pleasure to present before you this book We have made an attempt to provide chapter wise questions asked in JEE Main 2021 all 26 sets of February March July and August attempts with solutions Solutions to the questions are not just sketch rather have been written in such a manner that the students will be able to under the application of concept and can answer some other related questions too We firmly believe that the book in this form will definitely help a genuine hardworking student We have tried our best to keep errors out of this book Comment and criticism from readers will be highly appreciated and incorporated in the subsequent edition We wish to utilize the opportunity to place on record our special thanks to all team members of Content Development for their efforts to make this wonderful book

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