

Conversion factors:

Molar mass = #g/1 mole

Avogadro's number = 6.02×10^{23} atoms or molecules/1 mole

Molarity = #moles/1L

Name _____

Period _____

Stoichiometry Practice Problem Set

1. In the following reaction $4\text{Fe}(s) + 3\text{O}_2(g) \rightarrow 2\text{Fe}_2\text{O}_3(s)$, iron oxidizes to form rust.
 - a) If 40.0g of iron react, how many moles of Fe_2O_3 iron (III) oxide are produced?
 - b) If 20.0g of iron react, how many grams of Fe_2O_3 are produced?
 - c) If 10.0 moles of iron react, how many molecules of Fe_2O_3 are produced?
 - d) If 3.00×10^{23} atoms of iron react, how many moles of Fe_2O_3 are produced?
2. In the reaction for photosynthesis, $6\text{CO}_2(g) + 6\text{H}_2\text{O}(l) + \text{light energy} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6(aq) + 6\text{O}_2(g)$.
 - a) If 715g of CO_2 gas is consumed, how many grams of glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) are produced? The molar mass of glucose is 180.16g/mol.
 - b) How many molecules of carbon dioxide are required to produce 5.9×10^{24} molecules of oxygen gas?

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Morris Hein, Susan Arena

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A Stoichiometry Unit David Callaghan,2004 *Chemistry* John A. Olmsted,Gregory M. Williams,Robert Charles Burk,2016-01-14 Olmsted Burk is an introductory general chemistry text designed specifically with Canadian professors and students in mind A reorganized Table of Contents and inclusion of SI units IUPAC standards and Canadian content designed to engage and motivate readers distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant *Foundations of College Chemistry, Alternate* Morris Hein,Susan

Arena,2010-01-26 Learning the fundamentals of chemistry can be a difficult task to undertake for health professionals For over 35 years this book has helped them master the chemistry skills they need to succeed It provides them with clear and logical explanations of chemical concepts and problem solving They ll learn how to apply concepts with the help of worked out examples In addition Chemistry in Action features and conceptual questions checks brings together the understanding of chemistry and relates chemistry to things health professionals experience on a regular basis

Introduction to Chemistry, Laboratory Manual T. R. Dickson,1994-12-23 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and a chemical vision learning problem solving methods and utilizing group work and discussion activities These themes involve and engage the students in their own learning processes they are challenged to be active The presentation of topics has been altered to include a new chapter which introduces the students to scientific thinking and shows that chemistry involves interesting and relevant topics The reorganization presents many core concepts in the first five chapters preparing students for later chapters In addition the author has added vignettes throughout the chapters referring to health technology the environment and society as well as to specific tools of direct use to students

Journal of Research of the National Bureau of Standards United States. National Bureau of Standards,1972

Handbook of Zeolite Science and Technology Scott M. Auerbach,Kathleen A. Carrado,Prabir K. Dutta,2003-07-31 The Handbook of Zeolite Science and Technology offers effective analyses of salient cases selected expressly for their relevance to current and prospective research Presenting the principal theoretical and experimental underpinnings of zeolites this international effort is at once complete and forward looking combining fundamental

[The Software Finder](#) ,1983

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