

13.4 Practice Problems

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1. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then compressed to a volume of 0.500 L. Calculate the work done on the gas.
2. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then expanded to a volume of 2.00 L. Calculate the work done by the gas.
3. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then compressed to a volume of 0.500 L. Calculate the change in internal energy of the gas.
4. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then expanded to a volume of 2.00 L. Calculate the change in internal energy of the gas.
5. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then compressed to a volume of 0.500 L. Calculate the change in enthalpy of the gas.
6. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then expanded to a volume of 2.00 L. Calculate the change in enthalpy of the gas.
7. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then compressed to a volume of 0.500 L. Calculate the change in Gibbs free energy of the gas.
8. A gas is contained in a cylinder at a pressure of 1.00 atm and a volume of 1.00 L. The gas is then expanded to a volume of 2.00 L. Calculate the change in Gibbs free energy of the gas.

13 4 Practice Problems

Huangqi Zhang



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