

# 7-1 Review and Reinforcement

## Ionic Bonding

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

negative	noble gas	octet	electron configuration
positive	electrons	ions	polyatomic
crisscross	anion	cation	empirical formula

- An ionic bond is an attraction between oppositely charged —ions.
- Anions have a —negative charge.
- An atom becomes an ion by losing or gaining electrons.
- The octet rule states that atoms tend to gain, lose, or share electrons in order to acquire a full set of valence electrons.
- When sodium and chlorine form an ionic bond, both ions acquire the electron configuration of a(n) noble gas.
- The atoms that make up polyatomic ions are bonded together by covalent bonds.
- The crisscross method can be used to write the formula for an ionic compound.
- The empirical formula of a compound denotes the ratio of ions in the compound.

If the statement is true, write "true." If it is false, change the underlined word or words to make it true. Write your answer on the line.

<u>losing</u>	9. Calcium becomes a monatomic cation by <u>gaining</u> two electrons.
<u>true</u>	10. A cation has a positive charge.
<u>true</u>	11. A binary ionic compound contains only one kind of cation and one kind of anion.
<u>polyatomic</u>	12. Monatomic ions consist of more than one atom.
<u>seven</u>	13. The Lewis dot diagram for chlorine, a group 7A element, has six electrons.
<u>true (released)</u>	14. A great deal of energy is produced when an ionic compound is formed.

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**Thomas Fetsco, John McClure**

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scholars all over the world and has produced solid results This new text takes the next step It critically examines in one comprehensive volume the current prevalent approaches to cross cultural analysis at the level of nations that have been developed since Hofstede's work offering students and researchers the theoretical and practical advantages and potential pitfalls of each method The book is structured into four distinct parts Parts I and II focus on the main theoretical and statistical issues in cross cultural analysis using Hofstede's approach and the different research methods now associated with it Part II consists of presentations of all well known and some lesser known large scale cross cultural studies since Hofstede's work that have explained cross cultural variation in terms of dimensional models Part III summarizes the main conclusions to be drawn from the presentations in Part II and I explains how the proposed models have contributed to our practical understanding of cross cultural diversity

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**School Psychology Review**, 1998    CAS Review, 1975    Paramedic Care Vol 1 Irm Vol 1 Sup  
Bledsoe, Porter, 2005-03    **Neural Computation**, 2005    *Civil Engineering and Public Works Review*, 1963

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