

Assessment**Chapter Test A****Thermodynamics****MULTIPLE CHOICE**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. What accounts for an increase in the temperature of a gas that is kept at constant volume?
 - a. Energy has been removed as heat from the gas.
 - b. Energy has been added as heat to the gas.
 - c. Energy has been removed as work done by the gas.
 - d. Energy has been added as work done on the gas.

- _____ 2. An ideal gas system is maintained at a constant volume of 4 L. If the pressure is constant, how much work is done by the system?

a. 0 J	c. 8 J
b. 5 J	d. 30 J

- _____ 3. What thermodynamic process for an ideal gas system has an unchanging internal energy and a heat intake that corresponds to the value of the work done by the system?

a. isovolumetric	c. adiabatic
b. isobaric	d. isothermal

- _____ 4. Which thermodynamic process takes place when work is done on or by the system but no energy is transferred to or from the system as heat?

a. isovolumetric	c. adiabatic
b. isobaric	d. isothermal

- _____ 5. Which thermodynamic process takes place at a constant temperature so that the internal energy of a system remains unchanged?

a. isovolumetric	c. adiabatic
b. isobaric	d. isothermal

- _____ 6. According to the first law of thermodynamics, the difference between energy transferred to or from a system as heat and energy transferred to or from a system by work is equivalent to which of the following?

a. entropy change	c. volume change
b. internal energy change	d. pressure change

- _____ 7. An ideal gas system undergoes an adiabatic process in which it expands and does 20 J of work on its environment. What is the change in the system's internal energy?

a. -20 J	c. 0 J
b. -5 J	d. 20 J

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Michael Rajnik

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