

Assessment

Chapter Test A**Thermodynamics****MULTIPLE CHOICE**

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. What accounts for an increase in the temperature of a gas that is kept at constant volume?
- Energy has been removed as heat from the gas.
 - Energy has been added as heat to the gas.
 - Energy has been removed as work done by the gas.
 - Energy has been added as work done on the gas.
- _____ 2. An ideal gas system is maintained at a constant volume of 4 L. If the pressure is constant, how much work is done by the system?
- 0 J
 - 5 J
 - 8 J
 - 30 J
- _____ 3. What thermodynamic process for an ideal gas system has an unchanging internal energy and a heat intake that corresponds to the value of the work done by the system?
- isovolumetric
 - isobaric
 - adiabatic
 - isothermal
- _____ 4. Which thermodynamic process takes place when work is done on or by the system but no energy is transferred to or from the system as heat?
- isovolumetric
 - isobaric
 - adiabatic
 - isothermal
- _____ 5. Which thermodynamic process takes place at a constant temperature so that the internal energy of a system remains unchanged?
- isovolumetric
 - isobaric
 - adiabatic
 - isothermal
- _____ 6. According to the first law of thermodynamics, the difference between energy transferred to or from a system as heat and energy transferred to or from a system by work is equivalent to which of the following?
- entropy change
 - internal energy change
 - volume change
 - pressure change
- _____ 7. An ideal gas system undergoes an adiabatic process in which it expands and does 20 J of work on its environment. What is the change in the system's internal energy?
- 20 J
 - 5 J
 - 0 J
 - 20 J

10a Holt Physics Answers Temperature

Michael Rajnik

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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