



ACCOUNTING

Grade 12

Chapter 11

VAT

WORKSHEET

Compiled by
Mrs CW Brimecombe

Multiple Choice Quizz Of Accounting Grade 12

Robert R. Redfield

Multiple Choice Quizz Of Accounting Grade 12:

CBSE New Pattern Accountancy Class 12 for 2021-22 Exam (MCQs based book for Term 1) Harbinder Kaur, Abhishek Jain, 2021-09-10 1 This book deals with CBSE New Pattern Accountancy for Class 12 2 It is divided into 2 Parts as per Term 1 Syllabus 3 Quick Revision Notes covering all the Topics of the chapter 4 Carries all types of Multiple Choice Questions MCQs 5 Detailed Explanation for all types of questions 6 3 practice papers based on entire Term 1 Syllabus with OMR Sheet With the introduction of new exam pattern CBSE has introduced 2 Term Examination Policy where Term 1 deals with MCQ based questions while Term 2 Consists of Subjective Questions Introducing Arihant's CBSE New Pattern Series the first of its kind providing the complete emphasize on Multiple Choice Questions which are designated in TERM 1 of each subject from Class 9th to 12th Serving as a new preparatory guide here's presenting the all new edition of CBSE New Pattern Accountancy for Class 12 Term 1 that is designed to cover all the Term I chapters as per rationalized syllabus in a Complete Assertion Reasoning Based MCQs and Cased MCQs for the overall preparation Detailed Explanations of the selected questions help students to get the pattern and questions as well Lastly 3 Practice Questions are provided for the revision of the concepts TOC Part A Accounting for Partnership Firm Fundamentals Accounting for Partnership Firm Goodwill Reconstitution of Partnership Firm Change in the Profit Sharing Ratio Reconstitution of Partnership Firm Admission of a Partner Accounting of Share Capital Financial Statement of a Company Financial Statement Analysis Accounting Ratios Practice Papers 1 3

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web objective to apply stoichiometric principles to calculate the appropriate chemical equation for the reaction between iron metal and a solution of copper i sulfate introduction in the form of finely split powder copper metal is produced as a byproduct of the process as iron ions the ion meta dissolves

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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a

piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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web in this experiment we will use stoichiometric principles to deduce the appropriate equation for the reaction between metallic iron and a solution of copper ii sulfate this reaction produces metallic copper which is seen precipitating as a finely divided red powder

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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