

Multiple Choice

Select the alternative that best completes the statement or answers the question. What is most commonly observed with water?

1. How many water molecules form a hydrogen bond with a neighboring water molecule at typical temperatures?
  - a. 1
  - b. 2
  - c. 3
  - d. 4
2. At standard  $T_m$  (100°C), what factor determines whether a substance is a liquid?
  - a.  $\Delta H_{vap}$
  - b.  $\Delta H_{fus}$
  - c.  $\Delta H_{sub}$
  - d.  $\Delta H_{total}$
3. The enthalpy of fusion is the lowest for:
  - a. water at 0°C
  - b. ice at 0°C
  - c. liquid water at 0°C
  - d. solid water at 0°C
4. For a substance  $H_2O$ ,  $\Delta H_{vap} = 40.7 \text{ kJ mol}^{-1}$  and  $\Delta H_{fus} = 6.01 \text{ kJ mol}^{-1}$ . How many calories are spent during the phase transition associated with 1 mol of ice melting?
  - a. 60.7 kcal
  - b. 40.7 kcal
  - c. 6.01 kcal
  - d. 46.7 kcal
5. Water is the only substance that:
  - a. expands on freezing
  - b. contracts on freezing
  - c. contracts on melting
  - d. expands on melting
6. Water is the only liquid known that:
  - a. is colorless
  - b. is odorless
  - c. is tasteless
  - d. is all of the above
7. A substance's specific heat is  $100 \text{ J kg}^{-1} \text{ K}^{-1}$  at the triple point. It is placed in a Styrofoam container. The heater:
  - a. fails
  - b. works
  - c. stops
  - d. melts
8. For a substance  $H_2O$ ,  $\Delta H_{vap} = 40.7 \text{ kJ mol}^{-1}$  and  $\Delta H_{fus} = 6.01 \text{ kJ mol}^{-1}$ . How many grams of a substance (molar mass = 100 g/mol) are produced by heating?
  - a. 100 g
  - b. 1000 g
  - c. 10000 g
  - d. 100000 g
9. At standard  $T_m$  (100°C),  $\Delta H_{vap} = 40.7 \text{ kJ mol}^{-1}$  and  $\Delta H_{fus} = 6.01 \text{ kJ mol}^{-1}$ . What factor determines whether a substance is a liquid?
  - a.  $\Delta H_{vap}$
  - b.  $\Delta H_{fus}$
  - c.  $\Delta H_{sub}$
  - d.  $\Delta H_{total}$
10. The enthalpy of fusion is the lowest for:
  - a. water at 0°C
  - b. ice at 0°C
  - c. liquid water at 0°C
  - d. solid water at 0°C

# Multiple Choice Stoichiometry Test

**Richard C. Sweetland, Daniel J. Keyser**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the authors' names.

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countries Especially in the case of the school subject chemistry which is very unpopular on the one hand and occupies an exceptional position on the other hand due to its relevance to jobs and everyday life and most notably due to its importance for innovation capacity and problem solving we have to learn from each others educational systems

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**Media Review**, 1986      *The College Board Guide to the CLEP Examinations* College Entrance Examination Board, 1987      **Computer-assisted Instruction in Chemistry: General approach** Harry B. Mark, Hubert C. MacDonald, 1974

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