



ANSWER KEY

How many atoms of oxygen (O) are in the chemical formula below?

A 4
B 5
C 7
D 12



(d)

Carbon (C) and hydrogen (H) create covalent bonds to share electrons between them. What is the total number of electrons being shared between carbon and hydrogen in the molecule of propane shown below?

A 8
B 11

C 16
D 32



(c)

The diagram below represents the structural formula for several water molecules. How many molecules are represented?

A 3
B 5
C 10
D 15



(b)

Water molecules have some polarity in their structure. Using the diagram below, notice how the orbits of H atoms (positive) and O atoms (negative) meet. This property is the reason that

A water forms droplets easily
B water boils easily
C ice melts easily
D water has color



(a)

The chemical formula for the oxygen that we breathe is

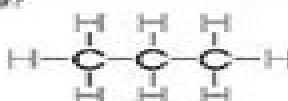
A O
B O₂
C O₃
D 2O



(b)

The diagram shown below is a structural formula for the gas propane. What would be the correct chemical formula for propane?

A C_3H_8
B CH_3
C $8\text{H}_3\text{C}$
D CH



(a)

The reason why the bonds between carbon (C) and hydrogen (H) are covalent is

A C and H do not have electrons to give away
B C and H are both nonmetals
C C and H are both metalloids
D C and H can bond only by sharing electrons



(d)

Explain the arrangement of the molecules in the diagram of water below.

A It is a lucky arrangement
B water molecules are sticky
C water molecules have polarity
D water molecules are magnetic



(c)

Since H atoms are slightly positive and the O atoms are slightly negative, molecules of water have polarity. This makes water a good

A solute
B refrigerant
C solvent
D magnet



(c)

How many molecules of oxygen are represented in the chemical formula below?

A 2
B 4
C 6
D 18



(b)

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M Mark

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