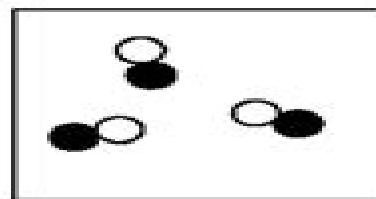
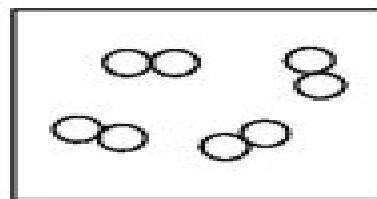
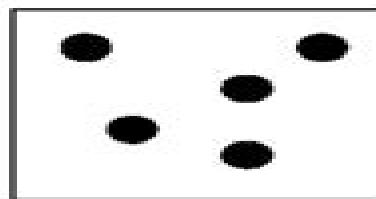


Elements, Compounds and Mixtures worksheet

1. Which of the boxes contains:

- Molecules of an element?
- Molecules of a compound?
- Atoms of an element?
- Do any of the boxes contain a mixture? Explain your answer.



2. Look at the following substances. Circle the elements and underline the compounds

a) Li b) N_2 c) CO d) Cl_2 e) Se f) NaCl

3. How many elements make up each compound?

a) HCl b) CaCO_3 c) HNO_3 d) H_2SO_4 e) CH_3OH

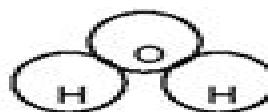
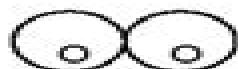
4. How many atoms are there in each of these molecules?

a) N_2 b) CO_2 c) O_3 d) CH_4 e) SO_2 f) CaH_4

5. Circle the molecule with the most atoms, in each pair:

a) HCl or POCl_3
b) NH_3 or HNO_3
c) CaHe or CaH_2OH
d) CH_3COCH_3 or CaHe

6. What is the formula for each molecules shown below?



Answers on the next page.

Atoms Elements And Compounds Answer Key

Didier Musso

Atoms Elements And Compounds Answer Key:

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Graybill,2015-10-01 This is the chapter slice What Are Compounds from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives Harcourt Science: Physical science [grade] 6, units E and F, teacher's ed ,2000

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Basic Chemistry William S. Seese,Guido H. Daub,1977

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