

Name: _____

Date/Period: _____

Atomic Number and Mass Number

Complete the following chart and answer the questions below.

<i>Element Name</i>	<i>Atomic Number</i>	<i>Number of Protons</i>	<i>Number of Neutrons</i>	<i>Mass Number</i>
carbon				12
	8		8	
hydrogen				1
		2		4
hydrogen			2	
nitrogen				14
			1	2
	92		146	
Potassium			20	
	11		12	
		47		108
Xenon			77	
			45	80
		24		52
			16	32
Gold				197
	76		114	

How are the *atomic number* and the *number of protons* related to each other?

How do the *number of protons*, *number of neutrons*, and the *mass number* relate to each other?

What is the *one thing* that determines the identity of an atom?

What sub atomic parts are equal to make a neutral atom?

Explain in your words how to calculate the mass of a sodium atom.

Draw and label three element symbols that are representative of the periodic table. It should have atomic #, mass #, name and element symbol.

Atomic Number Activity Answer

Philip L. Groisser



Atomic Number Activity Answer:

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