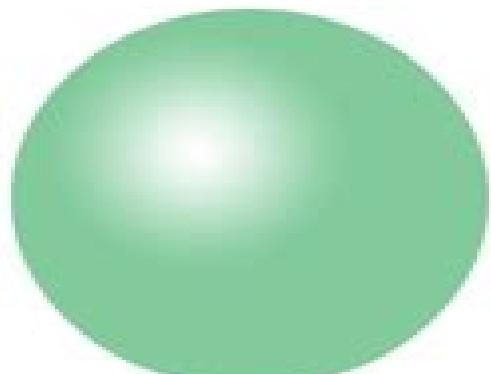


# Atomic Radius

VS

# Ionic Radius

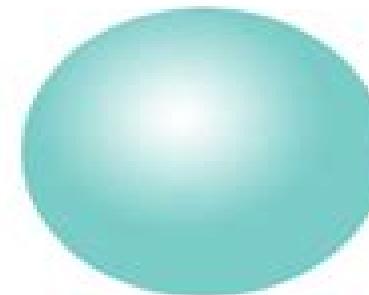
**Cation: Atomic Radius > Ionic Radius**



Na  
190 pm



$\text{Na}^+$   
116 pm



Mg  
145 pm

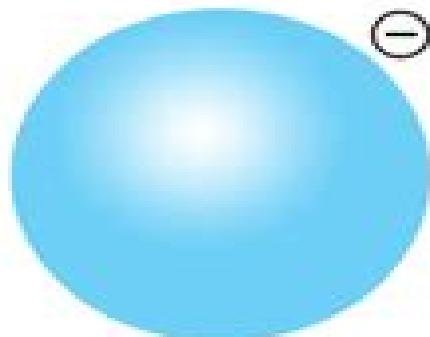


$\text{Mg}^{2+}$   
86 pm

**Anion: Atomic Radius < Ionic Radius**



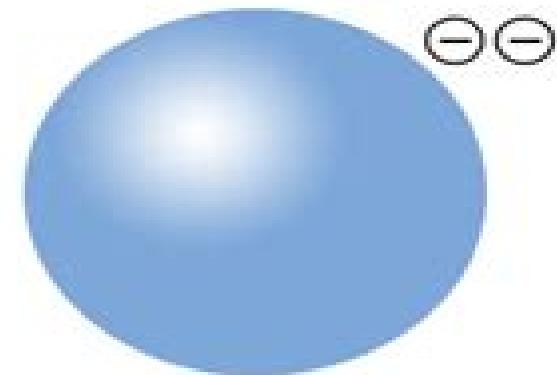
Cl  
79 pm



$\text{Cl}^-$   
167 pm



S  
88 pm



$\text{S}^{2-}$   
184 pm

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