



Atoms and Elements

Name: _____

Date: _____

chemical	mixtures	two-letter	brassine	bonds
chemical	cannot	gold	nitrogen	mercury
Mg	cannot	sulfur	different	compound
element	oxygen	hydrogen	impure	different
name	fluorine	distilling	carbon	Compounds
nitrogen	metals	hydrogen	elements	molecules
bonded	hydrogen	one	solids	atoms
atom	element	physical	single-letter	lattices
different	helium	filtering	oxygen	element
chemical	atom			

There is a huge number of _____ chemicals in our world. Some of these chemicals that will be familiar to you are glass, wood, water, salt, sugar, petrol and plastic.

These chemicals are all made out of extremely small particles called _____, which are joined together in _____ (groups of atoms) or _____ (networks).

All of the thousands of different chemicals that exist are made up from a limited number of different kinds of atoms.

An _____ is a simple chemical made up of only _____ type of atom. Pure gold is an element because it contains _____ atoms only.

An element _____ be broken down into simpler substances. Water can be broken down into _____ and _____ gases, so it is not an _____. Liquid mercury _____ be broken down into simpler substances, so it is an _____.

Most of the elements are _____. Eleven are gases (_____, _____, _____, nitrogen, oxygen, fluorine, neon, chlorine, etc.) and just two are liquids (_____ and _____).

Most of the elements are _____. Only 20 or so are non-metals (carbon, _____, _____, neon, phosphorus, sulfur, chlorine, iodine, etc.).

Each element has been given a distinct _____. Often the name was given to the element by the scientist.

Atoms Elements Compounds Cloze Activity

Robert R. Redfield

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Science Richard Moyer,2000 For grades 1-6 *The software catalog microcomputers* Menu (Firm) (Fort Collins, Colo.),1989 *The ETS Test Collection Catalog* Educational Testing Service. Test Collection,1993 The major source of information on the availability of standardized tests Wilson Library BulletinCovers commercially available standardized tests and hard to locate research instruments **Atoms, Molecules & Elements: What Are Molecules? Gr. 5-8** George Graybill,2015-10-01 This is the chapter slice What Are Molecules from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives **Atoms, Molecules & Elements: What Are Atoms? Gr. 5-8**

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