

Chemistry Atomic Structure Worksheet

① For the atomic structure shown below right, answer the following questions.

- Which element is the atom shown above? _____
- What is the chemical symbol of the element? _____
- What is the mass of the atom shown above? _____
- How many valence electrons are shown in the atom above? _____

② Which subatomic particle did J.J. Thomson discover in order to create his plum pudding model of the atom? _____

③ The nucleus was discovered by which scientist:

- Thomson
- Dalton
- Rutherford
- Democritus

④ What is the mass of Carbon-14: _____

⑤ What is the number of protons in Carbon-14: _____

⑥ What is the number of neutrons in Carbon-14: _____

⑦ What is the number of protons in the radioactive Uranium-238: _____

⑧ What is the number of neutrons in Sulfur-33: _____

⑨ Neutral atoms of the same element can differ in

- Mass number
- Atomic number
- Number of protons and electrons

⑩ The basic principles of atomic theory were first conceived by

- Avogadro
- Bohr
- Dalton



Atomic Structure Chapter Worksheet

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Dowd, Janice D. Hall, Marian P.
Hattaway**

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Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004

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