

# Acids & Bases: Assessment Review

Name \_\_\_\_\_

*Skill – The student will be able to identify the acid, base, conjugate acid, and conjugate base in an acid – base reaction.*

1. Write a chemical equation showing how  $\text{HCO}_3^-$  behaves as an acid when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

2. Write a chemical equation showing how ammonia ( $\text{NH}_3$ ) behaves as a base when dissolved in water. Label the acid, base, conjugate acid & conjugate base.

*Skill – The student will be able to calculate with the  $[\text{H}^+]$  or  $[\text{OH}^-]$  from the ion product ( $K_w$ ) or pH from pOH (a one-step pH block calculation).*

3. Recreate the calculation flowchart on the diagram provided. (Be careful...it's been flipped!)



4. What is  $[\text{OH}^-]$  if a solution has  $[\text{H}^+] = 7.1 \times 10^{-4} \text{ M}$ ? Is the solution acidic, basic, or neutral?

5. Calculate  $[\text{H}^+]$  in a solution if  $[\text{OH}^-] = 5.1 \times 10^{-11} \text{ M}$ . Is the solution acidic, basic, or neutral?

6. Calculate the  $[\text{H}^+]$  in a solution that shows a pH of 11.75. Is the solution acidic, basic, or neutral?

7. What is the pOH of a solution if  $[\text{OH}^-] = 5.0 \times 10^{-4} \text{ M}$ ? Is the solution acidic, basic, or neutral?

# Assessment Acids And Bases Answer Key

**RC Schank**



## **Assessment Acids And Bases Answer Key:**

*Fluids and Electrolytes* Catherine Paradiso, 1999 This book is designed to prepare students for classes or NCLEX by providing a comprehensive outline review of this particularly difficult area of study. *Fluids and Electrolytes* offers extensive self testing that includes answer keys with rationale for correct and incorrect responses as well as a comprehensive outline review and chapter study questions and a comprehensive examination at the end of the book. Each question is categorized according to the components of the National Council of State Boards of Nursing Licensing Examinations NCLEX. Includes FREE disk containing more NCLEX questions.

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**Fluid, Electrolyte, and Acid-Base Imbalances** Allison

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## **Table of Contents Assessment Acids And Bases Answer Key**

1. Understanding the eBook Assessment Acids And Bases Answer Key
  - The Rise of Digital Reading Assessment Acids And Bases Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Assessment Acids And Bases Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Assessment Acids And Bases Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Assessment Acids And Bases Answer Key

- Personalized Recommendations
  - Assessment Acids And Bases Answer Key User Reviews and Ratings
  - Assessment Acids And Bases Answer Key and Bestseller Lists
5. Accessing Assessment Acids And Bases Answer Key Free and Paid eBooks
    - Assessment Acids And Bases Answer Key Public Domain eBooks
    - Assessment Acids And Bases Answer Key eBook Subscription Services
    - Assessment Acids And Bases Answer Key Budget-Friendly Options
  6. Navigating Assessment Acids And Bases Answer Key eBook Formats
    - ePub, PDF, MOBI, and More
    - Assessment Acids And Bases Answer Key Compatibility with Devices
    - Assessment Acids And Bases Answer Key Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Assessment Acids And Bases Answer Key
    - Highlighting and Note-Taking Assessment Acids And Bases Answer Key
    - Interactive Elements Assessment Acids And Bases Answer Key
  8. Staying Engaged with Assessment Acids And Bases Answer Key
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Assessment Acids And Bases Answer Key
  9. Balancing eBooks and Physical Books Assessment Acids And Bases Answer Key
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Assessment Acids And Bases Answer Key
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Assessment Acids And Bases Answer Key
    - Setting Reading Goals Assessment Acids And Bases Answer Key
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Assessment Acids And Bases Answer Key



- Fact-Checking eBook Content of Assessment Acids And Bases Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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