

6(d)(i)	<p>Any one from</p> <ul style="list-style-type: none"> <li>• to sterilise / disinfect water</li> <li>• to destroy / kill microorganisms / bacteria / microbes / pathogens</li> </ul>	1	<p>Ignore 'to clean water'.</p> <p>Ignore 'water purification' and 'germs'.</p> <p>Credit 'remove bacteria etc' / prevent algae'.</p>
6(d)(ii)	<p>The (health) benefit outweighs the risk</p> <p>OR</p> <p>a clear statement that once it has done its job, little of it remains</p> <p>OR</p> <p>used in (very) dilute concentrations / small amounts / low doses</p>	1	
6(d)(iii)	<p><math>\text{Cl}_2 + \text{H}_2\text{O} \longrightarrow \text{HClO} + \text{HCl}</math></p> <p>OR</p> <p><math>\text{Cl}_2 + \text{H}_2\text{O} \longrightarrow 2\text{H}^+ + \text{ClO}^- + \text{Cl}^-</math></p> <p>OR</p> <p><math>2\text{Cl}_2 + 2\text{H}_2\text{O} \longrightarrow 4\text{HO} + \text{O}_2</math></p>	1	<p>Credit <math>\text{HClO}</math> or <math>\text{ClOH}</math></p> <p>Or multiples.</p> <p>Credit other ionic or mixed representations.</p> <p>Ignore state symbols.</p>
6(e)	<p>In either order - Both required for one mark only</p> <p><math>\text{NaClO}</math> (OR <math>\text{NaOCl}</math>) <u>and</u> <math>\text{NaCl}</math></p>	1	<p>Credit correct ionic formulas.</p> <p>Give credit for answers in equations unless contradicted.</p>

# Accn1 Jun 2014 Paper

**Nihon Butsuri Gakkai**

**Accn1 Jun 2014 Paper:**

*Journal of the Physical Society of Japan* Nihon Butsuri Gakkai, 1990

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