

Chapter 2. Chemistry Comes Alive (Mastering A&P – Chapter Quiz)

- When atoms or molecules combine to form larger, more complex molecules, the process is known as _____.
 - Catabolism
 - Decomposition
 - Anabolism
 - Exchange
- A chemical reaction in which bonds are broken is _____.
 - Catabolic and endergonic
 - Anabolic and endergonic
 - Catabolic and exergonic
 - Anabolic and exergonic
- Which of the following factors would speed up the rate of a chemical reaction?
 - Lower temperature
 - Removal of the biological catalysts
 - A high concentration of reagents
 - Large particles
- Which of the following is considered the universal solvent?
 - Plasma
 - Blood
 - Intracellular fluid
 - Water
- In hydrolysis, _____.
 - A molecule of water is removed when proteins are synthesized from smaller molecules
 - Electrons are lost from a reactant
 - A molecule of water is added for each bond broken
 - A reactant accepts electrons
- All electrolytes are _____.
 - Polar molecules
 - Acids
 - Non-polar molecules
 - Ions
- pH is a concentration unit used to measure the concentration of _____.
 - Hydroxyl ions in a solution
 - Anions in a solution
 - Isotopes in a solution
 - Hydrogen ions in a solution

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