

# Acids and Bases

## Neutralization Reactions Lab



### Reading, Lab, and Practice Bundle (Grades 6-9)

#### Acids and Bases

##### What is a Neutralization

A neutralization reaction is when an acid reacts with a base to form a salt and water.

The type of chemical reaction important in everyday life – it involves an upset stomach or balancing acid pH in gardens.

##### What Are Acids and Bases

Acids are substances that can taste sour and can have a low pH.

Examples: lemon juice, vinegar.

#### Matching Vocabulary

Directions: Match each word on the left with the correct definition or example on the right. Write the correct letter next to each number.

(1) A substance with a pH greater than 7 that produces OH <sup>-</sup> ions.	(10)
(2) The product of an acid and base reaction, often found in the form of NaCl.	(2)
(3) A scale from 0 to 14 used to measure how acidic or basic a substance is.	(3)
(4) A chemical that changes color to show if a substance is acidic or basic.	(4)
(5) A substance with a pH less than 7 that produces H <sup>+</sup> ions.	(5)
(6) A reaction between an acid and a base to form water and salt.	(6)

NAME: \_\_\_\_\_ DATE: \_\_\_\_\_

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(6) A reaction between an acid and a base to form water and salt.	(6)

28 Pages Printable & Easy to Use Includes Answer Key

# 19 4 Neutralization Reactions Section Review Answer Key

**Fred M. Dewey**

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