

Acids and Bases

Neutralization Reactions Lab



Reading, Lab, and Practice Bundle (Grades 6-9)

Acids and Bases

What is a Neutralization


A neutralization reaction is when an acid reacts with a base to form a salt and water.

This type of chemical reaction is important in everyday life – for treating an upset stomach or for balancing soil pH in gardens.


What Are Acids and Bases

Acids are substances that taste sour and can have a low pH.

Examples: lemon juice.




Matching Vocabulary



Directions: Match each word on the left with the correct definition. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces OH^- ions. ()
- 2) The product of an acid and base reaction, often found in the form of NaCl . ()
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is. ()
- 4) A chemical that changes color to show if a substance is acidic or basic. ()
- 5) A substance with a pH less than 7 that produces H^+ ions. ()
- 6) A reaction between an acid and a base to form water and salt. ()

Matching Vocabulary



NAME: _____

DATE: _____

Directions: Match each word on the left with the correct definition or example on the right. Write the correct letter next to each number.

- 1) A substance with a pH greater than 7 that produces OH^- ions. ()
- 2) The product of an acid and base reaction, often found in the form of NaCl . ()
- 3) A scale from 0 to 14 used to measure how acidic or basic a substance is. ()
- 4) A chemical that changes color to show if a substance is acidic or basic. ()
- 5) A substance with a pH less than 7 that produces H^+ ions. ()
- 6) A reaction between an acid and a base to form water and salt. ()

Acid


Base

pH scale

Neutralization

Salt

Indicator



28 Pages Printable & Easy to Use Includes Answer Key

19 4 Neutralization Reactions Section Review Answer Key

Fred M. Dewey



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