

Acids and Bases Problems

1. Rank the following solutions in order of increasing acidity (from lowest to highest):

Solution A with $\text{pH} = 6.50$

Solution C with $[\text{OH}^-] = 6.0 \times 10^{-10} \text{ M}$

Solution B with $[\text{H}_3\text{O}^+] = 3.5 \times 10^{-5} \text{ M}$

Solution D with $\text{pH} = 5.85$

_____ < _____ < _____ < _____

2. Complete the following table.

Acid	Conjugate Base
HS^-	
	ClO_4^-
H_3PO_4	
	HClO_3^-

3. Write the acid-base reactions with water for each of the following acids. Label the acid, base, conjugate acid, and conjugate base.

i) $[\text{Co}(\text{NH}_3)_5(\text{OH}_2)]^{3+}$

ii) HSO_4^-

iii) CH_3OH

iv) H_2PO_4^-

19 3 Practice Problems Chemistry Answers Acid And Bases

Neale Taylor, Angela Stubbs, Robert Stokes



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