

# 14

## THE BEHAVIOR OF GASES

### Chapter Quiz

Indicate whether the statement is true or false.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.
2. According to Boyle's law, if the pressure on a sample of gas is doubled, the volume of the gas is halved.
3. According to Charles's law, the volume of a gas is directly proportional to the absolute temperature.
4. The volume of a gas is directly proportional to the absolute temperature.
5. According to Charles's law, if the temperature of a gas is doubled, the volume of the gas is also doubled.
6. According to Charles's law, if the temperature of a gas is halved, the volume of the gas is also halved.
7. According to Charles's law, if the temperature of a gas is doubled, the volume of the gas is also doubled.
8. According to Charles's law, if the temperature of a gas is halved, the volume of the gas is also halved.
9. According to Charles's law, if the temperature of a gas is doubled, the volume of the gas is also doubled.
10. According to Charles's law, if the temperature of a gas is halved, the volume of the gas is also halved.

Indicate whether the statement is true or false.

1. A gas sample is held at a constant volume. If the pressure is doubled, the temperature is also doubled.

Indicate whether the statement is true or false.

1. A gas sample is held at a constant volume. If the pressure is doubled, the temperature is also doubled.

Indicate whether the statement is true or false.

1. A gas sample is held at a constant volume. If the pressure is doubled, the temperature is also doubled.

1. Boyle's law states that, at constant temperature, the volume of a fixed amount of gas is inversely proportional to the pressure.	True
2. According to Boyle's law, if the pressure on a sample of gas is doubled, the volume of the gas is halved.	True
3. According to Charles's law, the volume of a gas is directly proportional to the absolute temperature.	True
4. The volume of a gas is directly proportional to the absolute temperature.	True
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6. According to Charles's law, if the temperature of a gas is halved, the volume of the gas is also halved.	True
7. According to Charles's law, if the temperature of a gas is doubled, the volume of the gas is also doubled.	True
8. According to Charles's law, if the temperature of a gas is halved, the volume of the gas is also halved.	True
9. According to Charles's law, if the temperature of a gas is doubled, the volume of the gas is also doubled.	True
10. According to Charles's law, if the temperature of a gas is halved, the volume of the gas is also halved.	True

True

True

True

# 14 The Behavior Of Gases Chapter Quiz

**N Colangelo**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author's name.

## 14 The Behavior Of Gases Chapter Quiz:

**Instructor's Manual to Accompany Introductory Applied Physics, Third Edition** Norman C. Harris, 1972 **Test Book to Accompany Miller and Schroeer College Physics Sixth Edition** Dietrich Schroeer, Franklin D. Miller, Jr., Robert W. Stanley, 1987 New edition of a standard college physics textbook **Study and Quiz Outline** William Lloyd Evans, 1923

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field environment The 1965 Cambric event a small detonation in the saturated zone of the Valley Fill alluvial aquifer at NTS has been studied extensively to improve the understanding of groundwater flow and radionuclide migration The 1974 field investigation of the Cambric event included 1 a re entry boring into the original detonation cavity and 2 the installation of a pumping well 91 meters away from the cavity to draw mobile radionuclides such as tritium and krypton 85 This satellite well sampled regularly over 16 years produced some unexpected results Because of the supposed ideality of tritium and krypton as tracers their arrival was expected to be concurrent however krypton 85 arrival was delayed compared to tritium More important is the fact that while over 90% of the decay corrected tritium from the original blast was drawn out through the well less than half of the krypton was recovered This work presents a hypothesis for understanding these unexpected results based on the site hydrogeology and underground detonation phenomenology Carbon dioxide evolved from caliche during the detonation appears to have caused the upward migration of krypton 85 and other gaseous radionuclides through the collapsing chimney and into the unsaturated zone above In the process krypton became distributed throughout the cavity and chimney while tritium remained immediately around the cavity This post detonation movement of krypton 85 provides one explanation then of the unexpected elution curve results To demonstrate the validity of this hypothesis a variety of numerical simulations were performed using the finite difference code TRACR3D Calibrating two unknown hydraulic conductivity parameters on the tritium field data it was shown that krypton 85 may indeed have been distributed throughout the chimney as suggested by the hypothesis Abstract      Introductory Chemistry Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking      *Code of Federal Regulations* ,1992 Special edition of the Federal Register containing a codification of documents of general applicability and future effect as of July 1 with ancillaries

*Butane-propane News* ,1945      **Corrosion Tests and Standards** Robert Baboian, 2005      **Fuels and Lubricants Handbook** ,      Journal of Research of the National Bureau of Standards United States. National Bureau of Standards, 1976

**NBS Publications Newsletter** ,1964 A newsletter for librarians documentalists and science information specialists

**Guidelines for Engineering Design for Process Safety** ,1993 Inherently safer plants begin with the initial design Here is where integrity and reliability can be built in at the lowest cost and with maximum effectiveness This book focuses on process safety issues in the design of chemical petrochemical and hydrocarbon processing facilities It discusses how to select designs that can prevent or mitigate the release of flammable or toxic materials which could lead to a fire explosion or environmental damage All engineers on the design team the process hazard analysis team and those who make basic

decisions on plant design will benefit from its comprehensive coverage its organization and the extensive references to literature codes and standards that accompany each chapter      **The Principles of Materials Selection for Engineering Design** P. L. Mangonon,1999 Introducing readers to the methodology of engineering design the book shows how materials selection comes into play during the design of a component or a structure and examines such engineering requirements as stress mode of loading corrosion and performance efficiencies of materials Readers are acquainted with the factors of costs and statutory requirements including environmental regulations and recycling and case studies are integrated throughout to illustrate the selection process      *Harcourt Science: Teacher's ed., life science units A and B* ,2005      **Scientific American** ,1896      *Forest and Stream* ,1893

## **14 The Behavior Of Gases Chapter Quiz** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**14 The Behavior Of Gases Chapter Quiz**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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