

PHYSICAL SCIENCE PRACTICE TEST WITH ANSWER KEY

1. A weak attraction between hydrogen atom in one molecule and an oxygen in other called _____.
A. covalent bond
B. Hydrogen bond
C. Ionic bond
D. Metallic bond
2. Which of the following substances is most likely a covalent compound?
A. CaCO_3
B. KNO_3
C. SO_2
D. NaOH
3. What is the element needed to bond with Na to create the most ionic bond?
A. F
B. Cl
C. Br
D. I
4. The boiling point of substance A is higher than that of substance B. What does this meaning?
A. Substance A is less viscous than substance B.
B. Substance A evaporates faster than substance B.
C. Substance B has greater surface tension than substance A.
D. Substance B has weaker intermolecular forces than substance A.
5. What is the intermolecular force of attraction between the nucleus of one atom and the electrons of a nearby atom?
A. Dipole-dipole force
B. Ion-dipole force
C. London dispersion force
D. Van der Waals force
6. Which of the following has the greatest number of atom?
A. $\text{Ca}(\text{OH})_2$
B. Al_2O_3

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