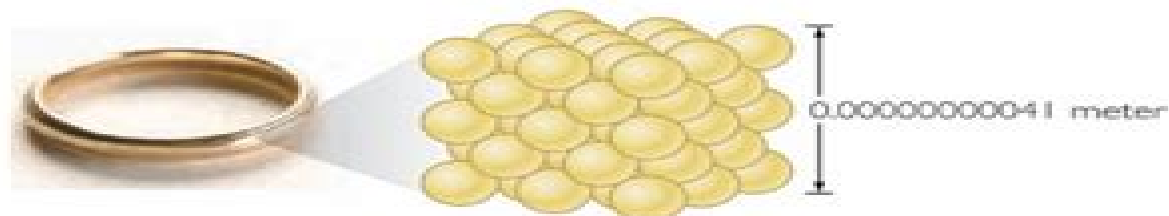


LESSON 11

Atomic Pudding Models of the Atom

Think About It

The drawing depicts a very tiny sample of gold taken from a gold ring.



The spheres in the cube of gold are so small that they cannot be seen. What are the spheres, and what does this drawing tell you about the element gold?

How are the smallest bits of matter described?

To answer this question, you will explore

- ① Atoms: Small Bits of Matter
- ② Models of the Atom
- ③ Simple Atomic Model

Exploring the Topic

① Atoms: Small Bits of Matter

Imagine you break a piece of matter in half, and then break it in half again and again. How many times can you do this? Can you keep going, getting ever smaller? Around 460 B.C.E., the Greek philosopher Democritus wondered the same thing. He thought that if he could just keep breaking matter in half he would eventually end up with the smallest bit of matter possible.

Answers Lesson 1 atomic Pudding

**Philip C. Keller, Raymond Chang, Jill L.
Keller**



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Student Solutions Manual for Chang's Chemistry Philip C. Keller, Raymond Chang, Jill L. Keller, 1988

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