

2000 AP[®] Chemistry Exam – Part I

Question 3.2

Consider atoms of the following elements. Assume that the atoms are in the ground state.

(A) Li	(E) Cl
(B) N	(F) S
(C) O	(G) Br
(D) F	(H) I

A

63.7%

11. The atom that contains exactly two unpaired electrons

C

78.9%

12. The atom that contains only one electron in the highest occupied energy level

Question 3.3 refers to the following molecules.

(A) H_2	(E) CO_2
(B) H_2O	(F) H_2O_2
(C) CH_4	(G) C_2H_4
(D) NH_3	(H) NH_3

D

74.7%

13. The molecule with only one double bond

E

64.9%

14. The molecule with the largest dipole moment

E

75.7%

15. The molecule that has trigonal pyramidal geometry

2002 Ap Chemistry Answer Key

H Kauffman



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