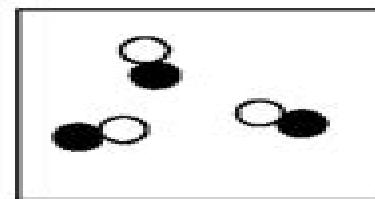
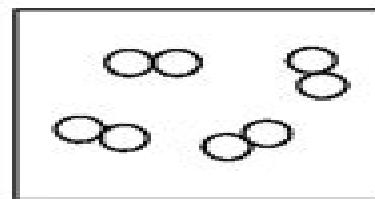
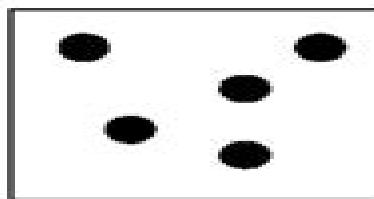


## Elements, Compounds and Mixtures worksheet

1. Which of the boxes contains:

- Molecules of an element?
- Molecules of a compound?
- Atoms of an element?
- Do any of the boxes contain a mixture? Explain your answer.



2. Look at the following substances. Circle the elements and underline the compounds

a) Li      b)  $\text{N}_2$       c) CO      d)  $\text{Cl}_2$       e) Se      f) NaCl

3. How many elements make up each compound?

a) HCl      b)  $\text{CaCO}_3$       c)  $\text{HNO}_3$       d)  $\text{H}_2\text{SO}_4$       e)  $\text{CH}_3\text{OH}$

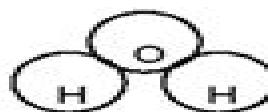
4. How many atoms are there in each of these molecules?

a)  $\text{N}_2$       b)  $\text{CO}_2$       c)  $\text{O}_3$       d)  $\text{CH}_4$       e)  $\text{SO}_2$       f)  $\text{CaH}_4$

5. Circle the molecule with the most atoms, in each pair:

a) HCl or  $\text{POCl}_3$   
b)  $\text{NH}_3$  or  $\text{HNO}_3$   
c)  $\text{CaHe}$  or  $\text{CaHeOH}$   
d)  $\text{CH}_3\text{COCH}_3$  or  $\text{CaHe}$

6. What is the formula for each molecules shown below:



Answers on the next page.

# Atoms Elements And Compounds Answer Key

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Basic Chemistry William S. Seese,Guido H. Daub,1977

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## **Table of Contents Atoms Elements And Compounds Answer Key**

1. Understanding the eBook Atoms Elements And Compounds Answer Key
  - The Rise of Digital Reading Atoms Elements And Compounds Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Atoms Elements And Compounds Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atoms Elements And Compounds Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atoms Elements And Compounds Answer Key
  - Personalized Recommendations
  - Atoms Elements And Compounds Answer Key User Reviews and Ratings
  - Atoms Elements And Compounds Answer Key and Bestseller Lists

5. Accessing Atoms Elements And Compounds Answer Key Free and Paid eBooks
  - Atoms Elements And Compounds Answer Key Public Domain eBooks
  - Atoms Elements And Compounds Answer Key eBook Subscription Services
  - Atoms Elements And Compounds Answer Key Budget-Friendly Options
6. Navigating Atoms Elements And Compounds Answer Key eBook Formats
  - ePUB, PDF, MOBI, and More
  - Atoms Elements And Compounds Answer Key Compatibility with Devices
  - Atoms Elements And Compounds Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Atoms Elements And Compounds Answer Key
  - Highlighting and Note-Taking Atoms Elements And Compounds Answer Key
  - Interactive Elements Atoms Elements And Compounds Answer Key
8. Staying Engaged with Atoms Elements And Compounds Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Atoms Elements And Compounds Answer Key
9. Balancing eBooks and Physical Books Atoms Elements And Compounds Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Atoms Elements And Compounds Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Atoms Elements And Compounds Answer Key
  - Setting Reading Goals Atoms Elements And Compounds Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atoms Elements And Compounds Answer Key
  - Fact-Checking eBook Content of Atoms Elements And Compounds Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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