

Atomic Structure and the Periodic Table Study Guide

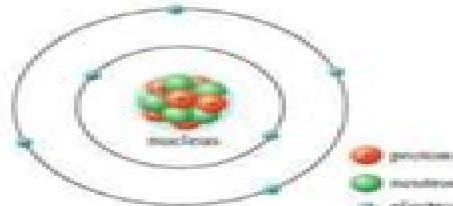
1. What is an atom?

Smallest particle of an element that retains the properties of that element.

2. Complete the following chart about subatomic particles.

Name	Charge	Symbol	Location	Relative Mass
Proton	+	p^+	Nucleus	1 amu
Electron	-	e^-	Electron cloud	0
neutron	0	n^0	nucleus	1 amu

3. Draw an atom and label the subatomic particles.

4. Which subatomic particle is used to determine the identity of an element?
protons5. Where is most of the mass of the atom? What particle(s) are in this area?
Nucleus; protons and neutrons

6. What is an atom called after it has gained or lost electrons? _____ion_____

7. What do you call an atom if it has more or less neutrons than normal? _____isotope_____

8. Label the information provided in the periodic table.

8	→	Atomic number
O	→	Symbol
Oxygen	→	Name
15.999	→	Atomic mass

9. From a box on the periodic table, how do you determine the number of protons, electrons, and neutrons?

1) # p^+ = _____ atomic number _____2) # e^- = _____ number of protons _____3) # n^0 = _____ atomic mass - atomic number _____

10. What does the atomic number represent? # protons in the nucleus

11. What does the atomic mass represent? # particles in the nucleus (protons & neutrons)

Atoms Study Guide Answers

Michael S. Matta

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its rediscoverers were biologists on the brink of discovering the chromosomal basis of heredity that is about to realize that mendel s heritable factors were carried on chromosomes

laws of inheritance flashcards quizlet - Dec 10 2022

study with quizlet and memorize flashcards containing terms like which of mendel s laws or principles explains that traits are passed from parents to offspring individually instead of as

chapter 8 holt biology mendel and heredity vocab - Apr 14 2023

verified answer biology assuming i that the two chromosomes in a homologous pair carry different alleles of some genes and

ii that no crossing over takes place how many