

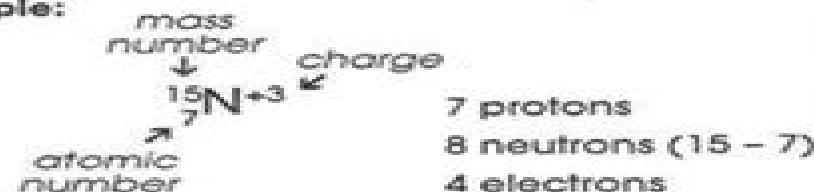
# ATOMIC STRUCTURE

Name \_\_\_\_\_

An atom is made up of protons and neutrons (both found in the nucleus) and electrons (in the surrounding electron cloud). The atomic number is equal to the number of protons. The mass number is equal to the number of protons plus neutrons. In a neutral atom, the number of protons equals the number of electrons. The charge on an ion indicates an imbalance between protons and electrons. Too many electrons produces a negative charge, too few, a positive charge.

This structure can be written as part of a chemical symbol.

Example:



Complete the following chart.

Element/ Ion	Atomic Number	Atomic Mass	Mass Number	Protons	Neutrons	Electrons
H						
H <sup>+</sup>						
$^{12}\text{C}$	6					
$^7\text{Li}^+$	3					
$^{35}\text{Cl}^-$	17					
$^{39}\text{K}$	19					
$^{24}\text{Mg}^{2+}$	12					
As <sup>3-</sup>						
Ag						
Ag <sup>+</sup>						
S <sup>2-</sup>						
U						

# Atomic Structure Worksheet Chemistry If8766

**John David Main Smith**

## Atomic Structure Worksheet Chemistry If8766:

Chemistry & Atomic Structure John David Main Smith,1924    Atomic and Molecular Structure Joel Chaffee,2011-01-15

Learn about the atom what it is the people responsible for helping us understand it and how it affects us in the world today

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*Chemistry & Atomic Structure, Etc* John David Main SMITH, 1924    **Atomic Structure and Chemical Reactions** Nevin Katz, 2007-05    Featuring the comic strip Dr. Birdley investigates    **The Structure of the Atom** Ernest Rutherford, 1913    Elementary Atomic Structure Gordon Kemble Woodgate, 1998    Atomic Spectra and Atomic Structure Gerhard Herzberg, 1937    **Atomic Structure and Chemistry** I. Hepburn, 1926\*    Chemistry and Atomic Structure John D. Main-Smith, 1924    Atomic Structure Laying the Foundation, 2008-07    **Atomic Structure** Laying the Foundation, 2008-07    Elementary Atomic Structure G.K.

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