

# Atomic Structure Worksheet

1 How many protons and neutrons are present in an atom of calcium (Mass number = 40)?

Protons = \_\_\_\_\_

Neutrons = \_\_\_\_\_

2 How many protons and neutrons are present in an atom of krypton (Mass number = 84)?

Protons = \_\_\_\_\_

Neutrons = \_\_\_\_\_

3 What is the name, symbol, and charge of an ion that contains 12 protons and 10 electrons?

Name = \_\_\_\_\_

Symbol = \_\_\_\_\_

Charge = \_\_\_\_\_

4 What is the name, symbol, and charge of an ion that contains 15 protons and 18 electrons?

Name = \_\_\_\_\_

Symbol = \_\_\_\_\_

Charge = \_\_\_\_\_

5 A neutral atom has 85 protons and 125 neutrons. Show calculations wherever necessary.

a What is its atomic number? \_\_\_\_\_

b What is its mass number? \_\_\_\_\_

c How many electrons does the atom have? \_\_\_\_\_

d What is the name of the atom? \_\_\_\_\_

6 Assuming all the atoms have neutral charge, complete the following table using your knowledge on atomic structure.

|        | Protons | Electrons | Neutrons | Mass number |
|--------|---------|-----------|----------|-------------|
| Atom 1 |         | 8         | 8        |             |
| Atom 2 |         |           | 14       | 28          |
| Atom 3 | 22      |           |          | 48          |
| Atom 4 |         | 47        |          | 107         |
| Atom 5 |         | 83        | 126      |             |
| Atom 6 | 87      |           |          | 223         |

# Atomic Structure Periodic Table Answers Key 2013

**M Plany**

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