

Name: _____
Date/Period: _____

Atomic Number and Mass Number

Complete the following chart and answer the questions below.

<i>Element Name</i>	<i>Atomic Number</i>	<i>Number of Protons</i>	<i>Number of Neutrons</i>	<i>Mass Number</i>
carbon				12
	8		8	
hydrogen				1
		2		4
hydrogen			2	
nitrogen				14
			1	2
	92		146	
Potassium			20	
	11		12	
		47		108
Xenon			77	
			45	80
		24		52
			16	32
Gold				197
	76		114	

How are the *atomic number* and the *number of protons* related to each other?

How do the *number of protons*, *number of neutrons*, and the *mass number* relate to each other?

What is the *one thing* that determines the identity of an atom?

What sub atomic parts are equal to make a neutral atom?

Explain in your words how to calculate the mass of a sodium atom.

Draw and label three element symbols that are representative of the periodic table. It should have atomic #, mass #, name and element symbol.

Atomic Number Activity Answer

Jan Rader, Jan Ray

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author names.

Atomic Number Activity Answer:

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Chapter 1 Units and Measurements 3 Chapter 2 Motion in a Straight Line 10 Chapter 3 Motion in a Plane 17 Chapter 4 Laws of Motion 24 Chapter 5 Work Energy and Power 31 Chapter 6 System of Particles and Rotational Motion 39 Chapter 7 Gravitation 47 Chapter 8 Mechanical Properties of Solids 55 Chapter 9 Mechanical Properties of Fluids 62 Chapter 10 Thermal Properties of Matter 69 Chapter 11 Thermodynamics 76 Chapter 12 Kinetic Theory 83 Chapter 13 Oscillations 90 Chapter 14 Waves 98 Chapter 1 Electric Charges and Fields 105 Chapter 2 Electrostatic Potential and Capacitance 113 Chapter 3 Current Electricity 123 Chapter 4 Moving Charges and Magnetism 130 Chapter 5 Magnetism and Matter 138 Chapter 6 Electromagnetic Induction 145 Chapter 7 Alternating Current 153 Chapter 8 Electromagnetic Waves 159 Chapter 9 Ray Optics and Optical Instruments 166 Chapter 10 Wave Optics 174 Chapter 11 Dual Nature of Radiation and Matter 181 Chapter 12 Atoms 188 Chapter 13 Nuclei 194 Chapter 14 Semiconductor Electronics Materials Devices and Simple Circuits 200

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Table of Contents Atomic Number Activity Answer

1. Understanding the eBook Atomic Number Activity Answer
 - The Rise of Digital Reading Atomic Number Activity Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Number Activity Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Number Activity Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Number Activity Answer
 - Personalized Recommendations
 - Atomic Number Activity Answer User Reviews and Ratings
 - Atomic Number Activity Answer and Bestseller Lists

5. Accessing Atomic Number Activity Answer Free and Paid eBooks
 - Atomic Number Activity Answer Public Domain eBooks
 - Atomic Number Activity Answer eBook Subscription Services
 - Atomic Number Activity Answer Budget-Friendly Options
6. Navigating Atomic Number Activity Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Number Activity Answer Compatibility with Devices
 - Atomic Number Activity Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Number Activity Answer
 - Highlighting and Note-Taking Atomic Number Activity Answer
 - Interactive Elements Atomic Number Activity Answer
8. Staying Engaged with Atomic Number Activity Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Number Activity Answer
9. Balancing eBooks and Physical Books Atomic Number Activity Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Number Activity Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Number Activity Answer
 - Setting Reading Goals Atomic Number Activity Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Number Activity Answer
 - Fact-Checking eBook Content of Atomic Number Activity Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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