

Name: _____

Period: _____

Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:

number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



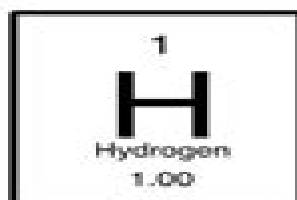
Protons:

7

Electrons:

7

Neutrons:

7

Protons:

1

Electrons:

1

Neutrons:

0

Protons:

8

Electrons:

8

Neutrons:

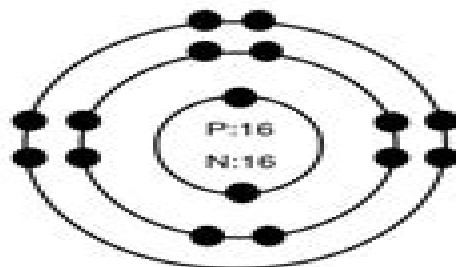
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BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
 1. The first energy level can hold 2 electrons
 2. The second energy level can hold 8 electrons
 3. The third energy level can hold 8 electrons

The electrons in the outermost energy level are called:

Valence Electrons

Element: SulfurNumber of Valence Electrons: 6

LEWIS DIAGRAMS

A lewis diagram shows only the symbol with the Valence electrons



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Tina Youngman



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Atomic Structure Workbook Tina Youngman, 2017

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Table of Contents Atomic Structure Worksheet Tesccc Answer

1. Understanding the eBook Atomic Structure Worksheet Tesccc Answer
 - The Rise of Digital Reading Atomic Structure Worksheet Tesccc Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Structure Worksheet Tesccc Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atomic Structure Worksheet Tesccc Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Structure Worksheet Tesccc Answer
 - Personalized Recommendations
 - Atomic Structure Worksheet Tesccc Answer User Reviews and Ratings
 - Atomic Structure Worksheet Tesccc Answer and Bestseller Lists

5. Accessing Atomic Structure Worksheet Tesccc Answer Free and Paid eBooks
 - Atomic Structure Worksheet Tesccc Answer Public Domain eBooks
 - Atomic Structure Worksheet Tesccc Answer eBook Subscription Services
 - Atomic Structure Worksheet Tesccc Answer Budget-Friendly Options
6. Navigating Atomic Structure Worksheet Tesccc Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Atomic Structure Worksheet Tesccc Answer Compatibility with Devices
 - Atomic Structure Worksheet Tesccc Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atomic Structure Worksheet Tesccc Answer
 - Highlighting and Note-Taking Atomic Structure Worksheet Tesccc Answer
 - Interactive Elements Atomic Structure Worksheet Tesccc Answer
8. Staying Engaged with Atomic Structure Worksheet Tesccc Answer
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atomic Structure Worksheet Tesccc Answer
9. Balancing eBooks and Physical Books Atomic Structure Worksheet Tesccc Answer
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atomic Structure Worksheet Tesccc Answer
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atomic Structure Worksheet Tesccc Answer
 - Setting Reading Goals Atomic Structure Worksheet Tesccc Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atomic Structure Worksheet Tesccc Answer
 - Fact-Checking eBook Content of Atomic Structure Worksheet Tesccc Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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