



# Atomic Dating Game

**Directions:** Atoms are lonely hearts that are constantly in search of partners to bring stability to their lives. Your job is to play matchmaker and make each atom stable by determining how many valence electrons each element needs and finding a partner that will complete the valence energy level.

Let's meet our first bachelor, Mr. Sodium. Mr. Sodium is very lustrous but has an explosive personality when he gets near water or oxygen. He is quite a lonely metal that, like all metals, is looking to lose a few electrons! Since he is highly reactive... be careful who you pick to be his date!

In the circle below, complete the shell model of Mr. Sodium.



11
<b>Na</b>
Sodium
22

Atomic #: 11  
Mass #: 22  
Protons: 11  
Neutrons: 12  
Electrons: 11



Let's meet our eligible bachelorettes:

Ms. Lithium is a highly reactive metal looking to lose electrons.

3
<b>Li</b>
Lithium
7

Atomic #: 3  
Mass #: 7  
Protons: 3  
Neutrons: 4  
Electrons: 3

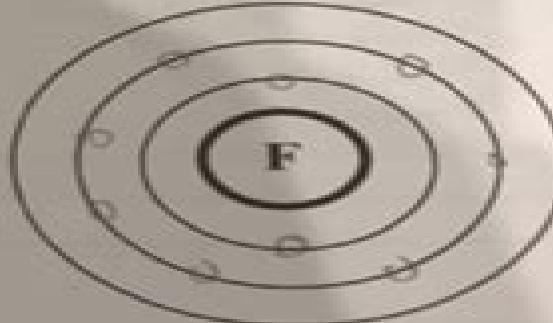


Mr. Sodium should date \_\_\_\_\_, because \_\_\_\_\_

Ms. Fluorine is a nonmetal looking to gain electrons.

9
<b>F</b>
Fluorine
19

Atomic #: 9  
Mass #: 19  
Protons: 9  
Neutrons: 10  
Electrons: 9



## Atom Dating Game Answers

**Paul Stanley Taylor**

## Atom Dating Game Answers:

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