

Activity Series Worksheet **KEY**

Use the activity series to determine if a reaction will occur. (A reaction will occur if the element **NOT** in the compound is higher on the activity series than the element that forms the same charged ion which is in the compound.) If a reaction will take place, predict the products and balance the reaction. If no reaction will occur, simply write, "no reaction". Notice the state of matter of your reactants and indicate the state of matter of your products. You may assume that all transition metals will form +2 ions.



Activity Series for Metals and Non-metals Single Replacement Reactions

Metal Name Symbol for ion	Decreasing activity
Lithium Li^+	reacts with water and acids
Potassium K^+	
Barium Ba^{2+}	
Strontium Sr^{2+}	
Calcium Ca^{2+}	
Sodium Na^+	
Magnesium Mg^{2+}	reacts with acids replacing hydrogen
Aluminum Al^{3+}	
Manganese Mn^{2+}	
Zinc Zn^{2+}	
Chromium Cr^{3+}	
Iron Fe^{2+}	
Cadmium Cd^{2+}	
Cobalt Co^{2+}	
Nickel Ni^{2+}	
Tin Sn^{2+}	
Lead Pb^{2+}	

Applying The Activity Series Answers

CL Gary



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