

SECTION - I

Q. 1. Select and write the most appropriate answers from the given alternatives for each sub-question :

[7]

- (i) The Arrhenius equation is _____.
- (a) $k = A e^{\frac{RT}{E_a}}$ (b) $A = k e^{\frac{-E_a}{RT}}$
- (c) $k = A e^{\frac{-RT}{E_a}}$ (d) $k = A e^{\frac{-K_a}{RT}}$
- (ii) If the enthalpy of vaporisation of water at 100°C is 186.5 J . mol⁻¹, the entropy of vaporization will be _____.
- (a) 4.0 J . K⁻¹. mol⁻¹ (b) 3.0 J . K⁻¹. mol⁻¹
- (c) 1.5 J . K⁻¹. mol⁻¹ (d) 0.5 J . K⁻¹. mol⁻¹
- (iii) The atomicity of sulphur in orthorhombic sulphur is _____.
- (a) 8 (b) 6 (c) 4 (d) 2
- (iv) The major binding force in diamond is _____.
- (a) Covalent bond (b) Ionic bond
- (c) Metallic bond (d) Co-ordinate covalent bond
- (v) The boiling point of water at high altitude is low, because _____.
- (a) the temperature is low. (b) the atmospheric pressure is low.
- (c) the temperature is high. (d) the atmospheric pressure is high.
- (vi) The molar conductivity of cation and anion of salt BA are 180 and 220 mhos respectively. The molar conductivity of salt BA at infinite dilution is _____.
- (a) 90 mhos.cm². (b) 110 mhos.cm².mol⁻¹
- (c) 200 mhos.cm².mol⁻¹ (d) 400 mhos.cm².mol⁻¹
- (vii) What is the process in which concentrated ore is reduced to the corresponding metal by heating at high temperature with a reducing agent?
- (a) Polling (b) Pyrometallurgy (c) Hydrometallurgy (d) Calcination

Q. 2. Answer any THREE of the following :

[9]

- (i) Describe anomalous behaviour of oxygen as compared with other elements of group 16 with reference to –
- (a) Magnetic property (b) Oxidation state (c) Hydrides
- (ii) What is the value of $\Delta S_{\text{surr.}}$ for the following reaction at 298 K –
- $$6\text{CO}_{2(g)} + 6\text{H}_2\text{O}_{(l)} \longrightarrow \text{C}_6\text{H}_{12}\text{O}_{6(s)} + 6\text{O}_{2(g)}$$
- Given that : $\Delta G^\circ = 2879 \text{ kJ mol}^{-1}$, $\Delta S = -210 \text{ JK}^{-1} \text{ mol}^{-1}$.
- (iii) Sucrose decomposes in acid solution to give glucose and fructose according to the first order rate law. The half life of the reaction is 3 hours. Calculate fraction of sucrose which will remain after 8 hours.
- (iv) A solution containing 0.73 g of camphor (molar mass 152 g . mol⁻¹) in 36.8 g of acetone (boiling point 56.3°C) boils at 56.55°C. A solution of 0.564 g of unknown compound in the same weight of acetone boils at 56.46°C. Calculate the molar mass of the unknown compound.

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Mike D. Adams



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Directory Medical Library Association, 1994

Large Meteorite Impacts and Planetary Evolution VI Wolf Uwe Reimold, Christian Koeberl, 2021-09-23 This volume contains a sizable suite of contributions dealing with regional impact records Australia Sweden impact craters and impactites early Archean impacts and geophysical

characteristics of impact structures shock metamorphic investigations post impact hydrothermalism and structural geology and morphometry of impact structures on Earth and Mars *Emerging Energy Alternatives for Sustainable Environment* D. P. Singh, Richa Kothari, V. V. Tyagi, 2019-02-14 Sustainability of environment is an emerging global issue at present Unsustainable or deteriorating environment is a matter of concern as it has threatened the survival of living creatures Recently climate change has been a matter of great concern at a global platform owing to imbalances in natural environment Increasing population has increased the demand for energy which has ultimately put pressure on natural resources and caused a paradigm shift from resource generation to exploitation *Emerging Energy Alternatives for Sustainable Environment* aims to address the role of sustainable technologies in energy generation options for clean environment It covers a wide spectrum of energy generation approaches with an emphasis on five key topics i renewable energy sources and recent advances ii emerging green technologies for sustainable development iii assessment of biomass for sustainable bioenergy production iv solid waste management and its potential for energy generation and v solar energy applications storage system and heat transfer This book provides essential and comprehensive knowledge of green energy technologies with different aspects for engineers technocrats and researchers working in the industry universities and research institutions The book is also very useful for undergraduate and graduate students of science and engineering who are keen to know about the development of renewable energy products and their corresponding processes Please note This volume is Co published with The Energy and Resources Institute Press New Delhi Taylor Francis does not sell or distribute the Hardback in India Pakistan Nepal Bhutan Bangladesh and Sri Lanka *Chemistry* Karen Enkelaar, Rosemary Peers, 2021 *Excel Success One HSC Chemistry* 2021 Edition contains Over 100 objective response and short answer questions from past HSC Papers 2001 2014 arranged into Module topics Four sample HSC Examination papers with actual questions from the 2015 2018 HSC papers All questions in these sample papers are on topics examinable in the new syllabus Over half of each paper s questions are actual past HSC questions and the others are written for the new topics in the updated syllabus Three Excel sample HSC Examination papers written to the new syllabus and based on the new HSC exam format The 2019 and 2020 HSC Examination papers Answers to all questions with distribution of marks for each question *Chemistry* Karen Enkelaar, Rosemary Peers, 2020 *Excel Success One HSC Chemistry* contains 2001 2014 2015 2018 past HSC questions and past 2019 HSC paper with detailed answers written by experienced HSC markers a topic index a mark maximizer guide and a glossary of key verbs This title helps you get the results you want by practising actual HSC papers and answering HSC level questions Directory of the Medical Library Association Medical Library Association, 1994 **The National Job Bank** , 2001 Australian Education Index , 1980 **Penn State Alumni Directory** , 2010 Success One HSC Chemistry 2014 Edit, 2013 EXCEL SUCCESS ONE HSC CHEMISTRY contains 2001 2003 and 2005 2013 past HSC questions with detailed answers written by experienced HSC markers a Topic Index a Mark Maximizer Guide and a Glossary of Key Verbs

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