

**General Instructions :**

Read the following instructions very carefully and strictly follow them :

- (i) This question paper comprises 15 questions. All questions are compulsory.
- (ii) This question paper is divided into three sections – **A**, **B** and **C**.
- (iii) **Section A** – Questions No. 1 to 7 are short answer type questions. Each question carries 2 marks.
- (iv) **Section B** – Questions No. 8 to 13 are also short answer type questions. Each question carries 3 marks.
- (v) **Section C** – Questions No. 14 and 15 are case-based questions. Each question carries 4 marks.
- (vi) Internal choices have been provided in some questions. Only one of the alternatives has to be attempted.

**SECTION A**

1. (a) Write the molecular formula of the following carbon compounds :  
 (i) Methane  
 (ii) Propane  
 (b) Carbon compounds have low melting and boiling points. Why ? 2
2. The electrons in the atoms of two elements X and Y are distributed in three shells having 1 and 7 electrons respectively in their outermost shells.  
 (a) Write the group numbers of these elements in the Modern Periodic Table.  
 (b) Write the molecular formula of the compound formed when X and Y combine with each other.  
 (c) Which of the two is electropositive ? 2
3. (a) Which of the following flowers will have higher possibility of self-pollination ?  
 Mustard, Papaya, Watermelon, Hibiscus  
 (b) List the two reproductive parts of a bisexual flower. 2
4. Which one of the two multicellular organisms — Spirogyra and Planaria reproduces by regeneration and why ? Give an example of any other organism which can also reproduce by the same process. 2

# 2013 Physical Science Question Paper Mid Year

**N Colangelo**

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